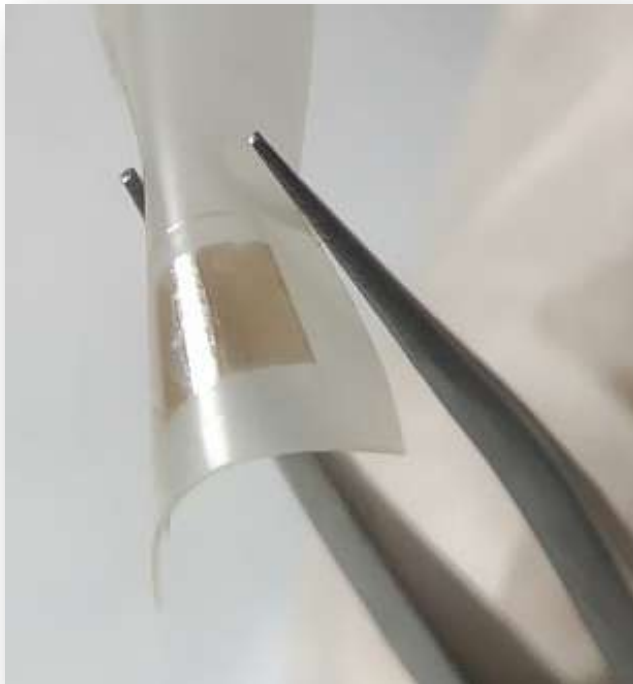


# Freestanding functional oxide membranes



Mariona Coll

Institut de Ciència de Materials de Barcelona

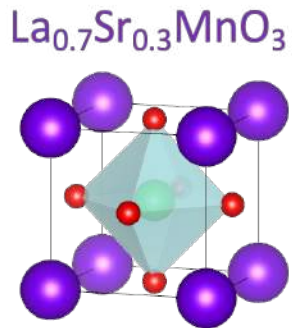
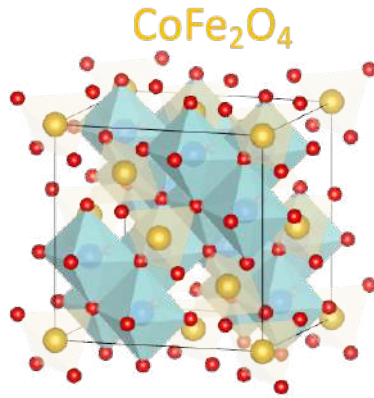
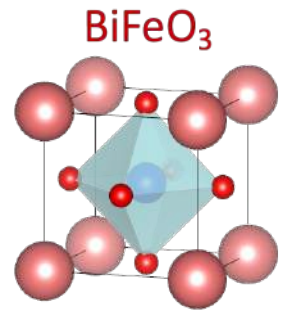
[mcoll@icmab.es](mailto:mcoll@icmab.es)

# Outline

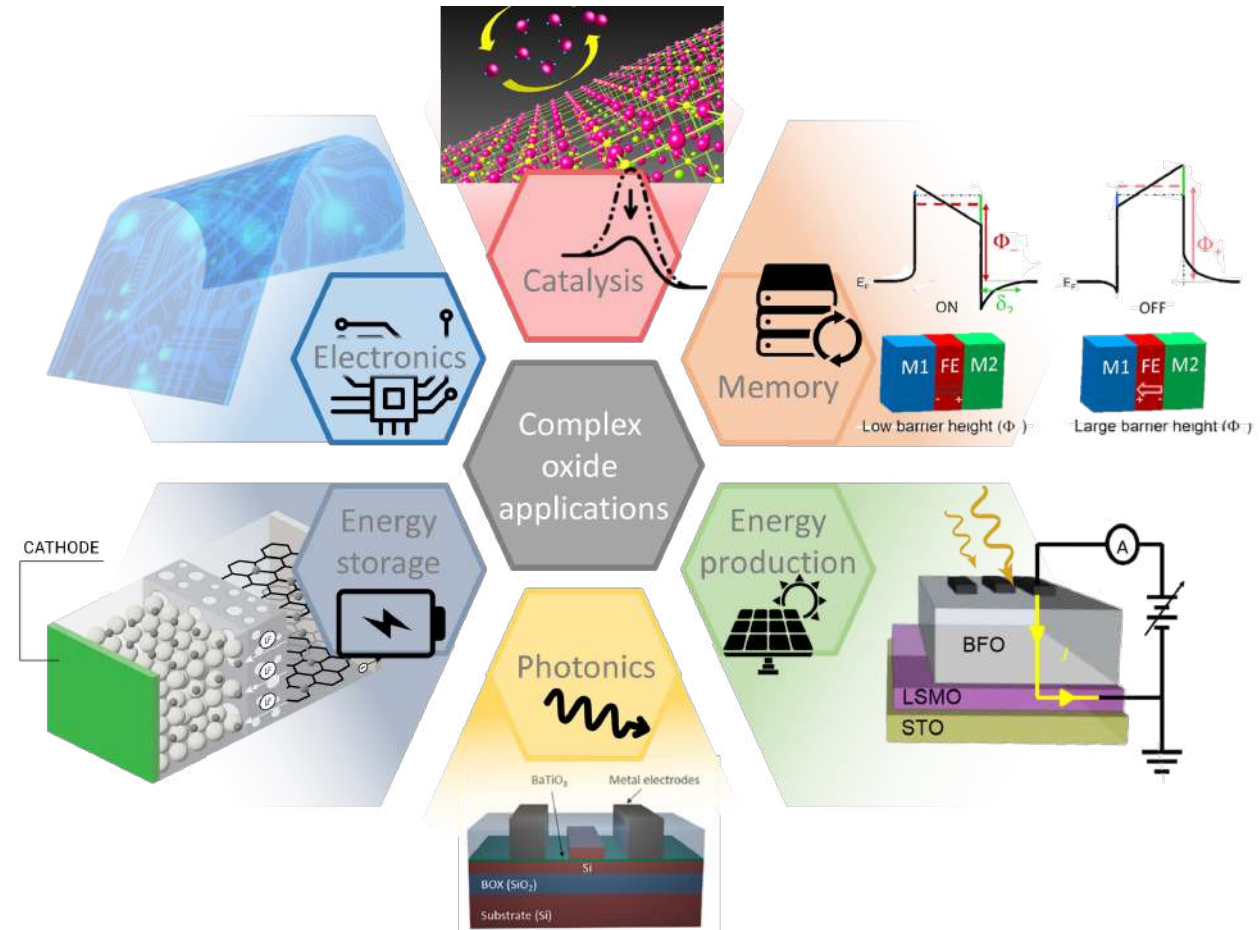
- Why free-standing oxides
- Preparation of free-standing oxides  
Sacrificial Layer
- Challenges in the preparation
- Outlook/ perspective

# Epitaxial complex oxides

- ✓ Broad variety of structures
- ✓ Composition versatility
- ✓ (chemical, thermal, mechanical) stability
- ✓ Unique physical, chemical properties

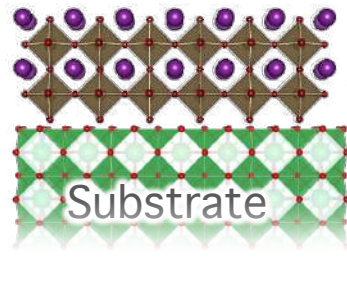


Photovoltaic  
 Ferroelectricity  
 Magnetism  
 Superconductivity  
 Metal-insulator transitions  
 ...



# Free-standing complex oxides

Epitaxial growth



Single-crystal  
Substrate

High Temperature  
treatment

New  
opportunities

Rigid

Expensive

Fragile

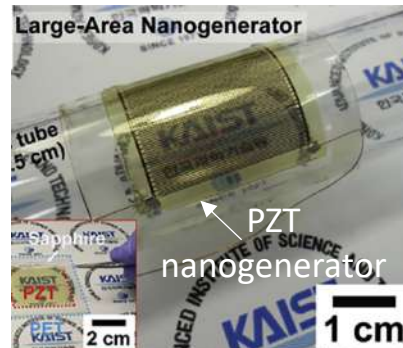
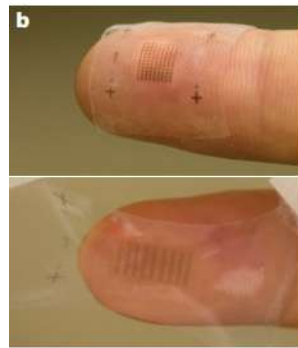
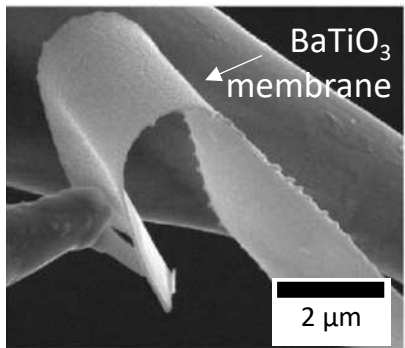
limited availability

limits substrate nature

Thermodynamic  
compatibility of films

# Epitaxial complex oxide membranes: new opportunities

- Integration into Si, polymers, 2D materials ...

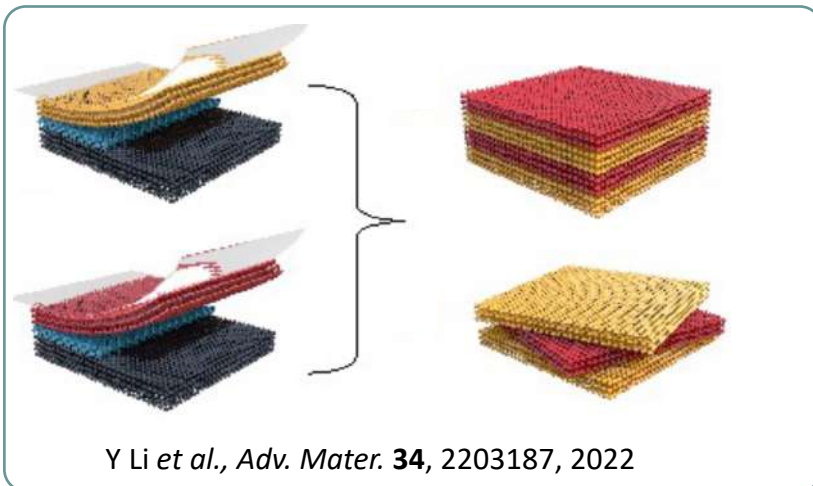


G Dong *et al.*, *Science* **366** (6464), 475–479, 2019

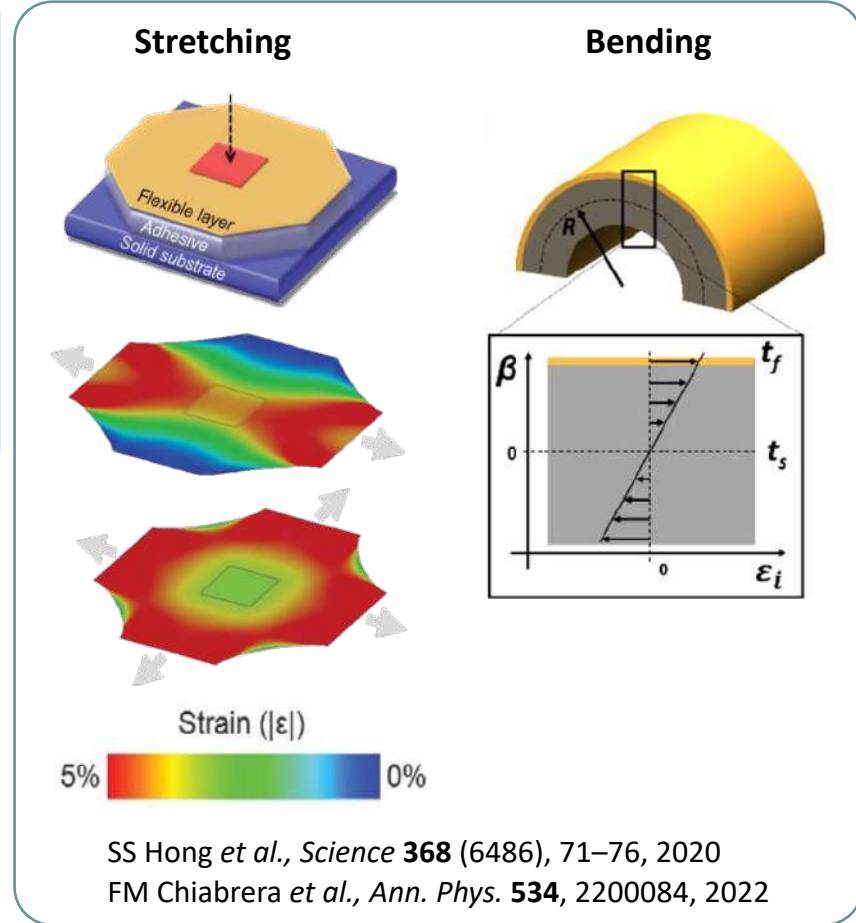
S Wang *et al.*, *Nature* **555** (7694), 83–88, 2018

K Park *et al.*, *Adv. Mater.* **26**, 2514–2520, 2014

- New phenomena at artificial interfaces



- Strain engineering



- Why free-standing oxides

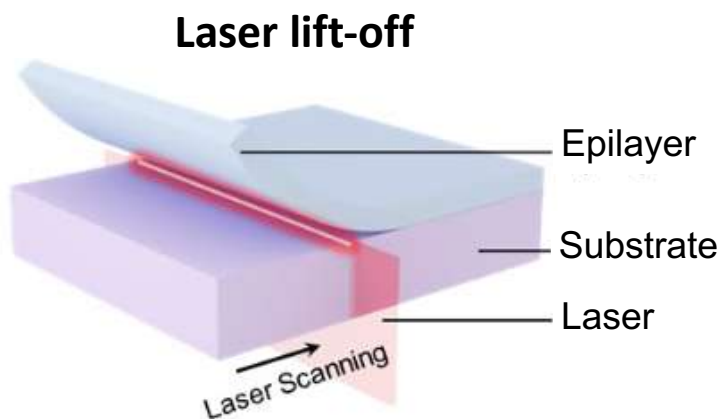
- **Preparation of free-standing oxides**

Sacrificial Layer

- Challenges in the preparation
- Outlook/ perspective

# Freestanding epitaxial complex oxides: How to detach them from the substrate?

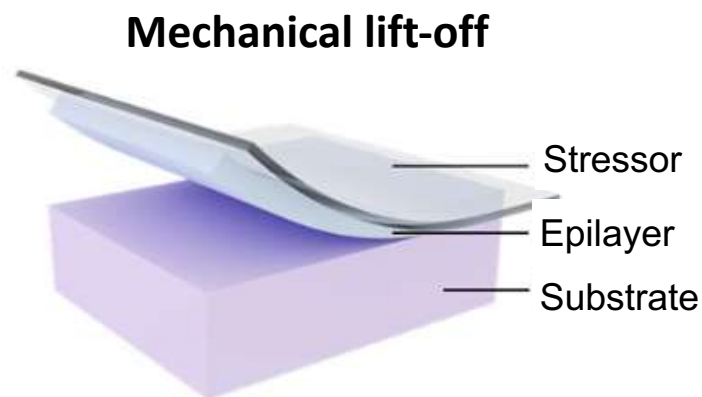
## PHYSICAL TECHNIQUES



Transparent substrate  
Bandgap of the film < bandgap laser

*Limited number of suitable films*  
*Rough surface*  
*No reusable substrate*

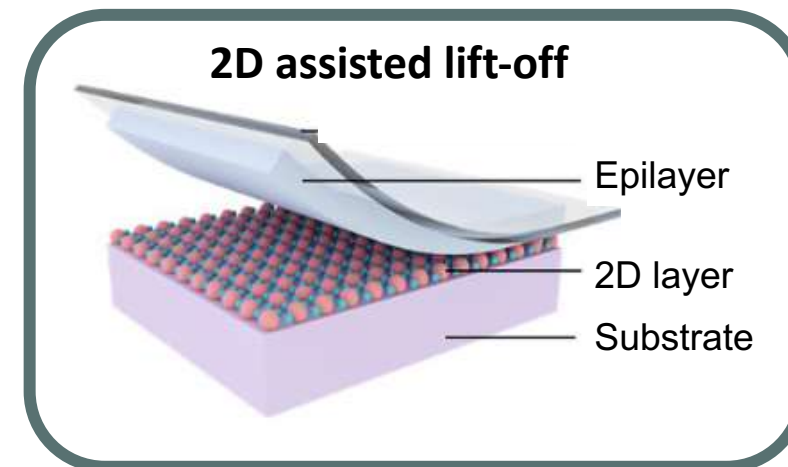
*Ex: PZT//sapphire-->PET*



Weak interface bonding  
Brute Force + stressor

*Materials with weak bonding*  
*Cracking*

*Ex: Ni/PMN-PT//SRO/STO*

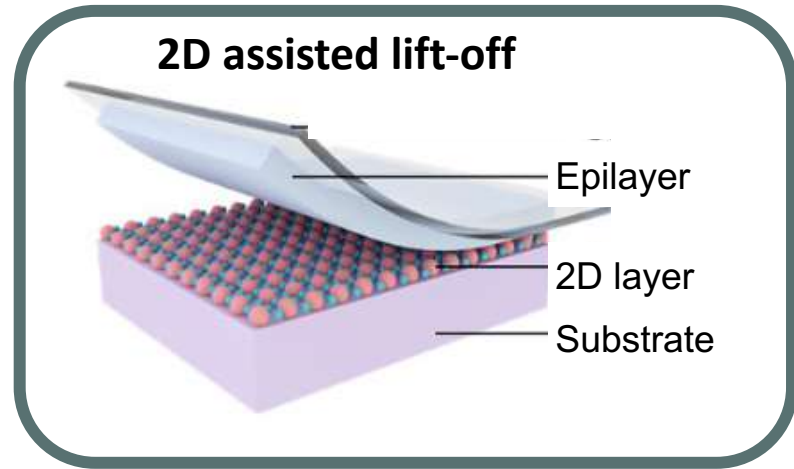


VdW 2D buffer layer  
Exfoliation yield > mechanical lift-off

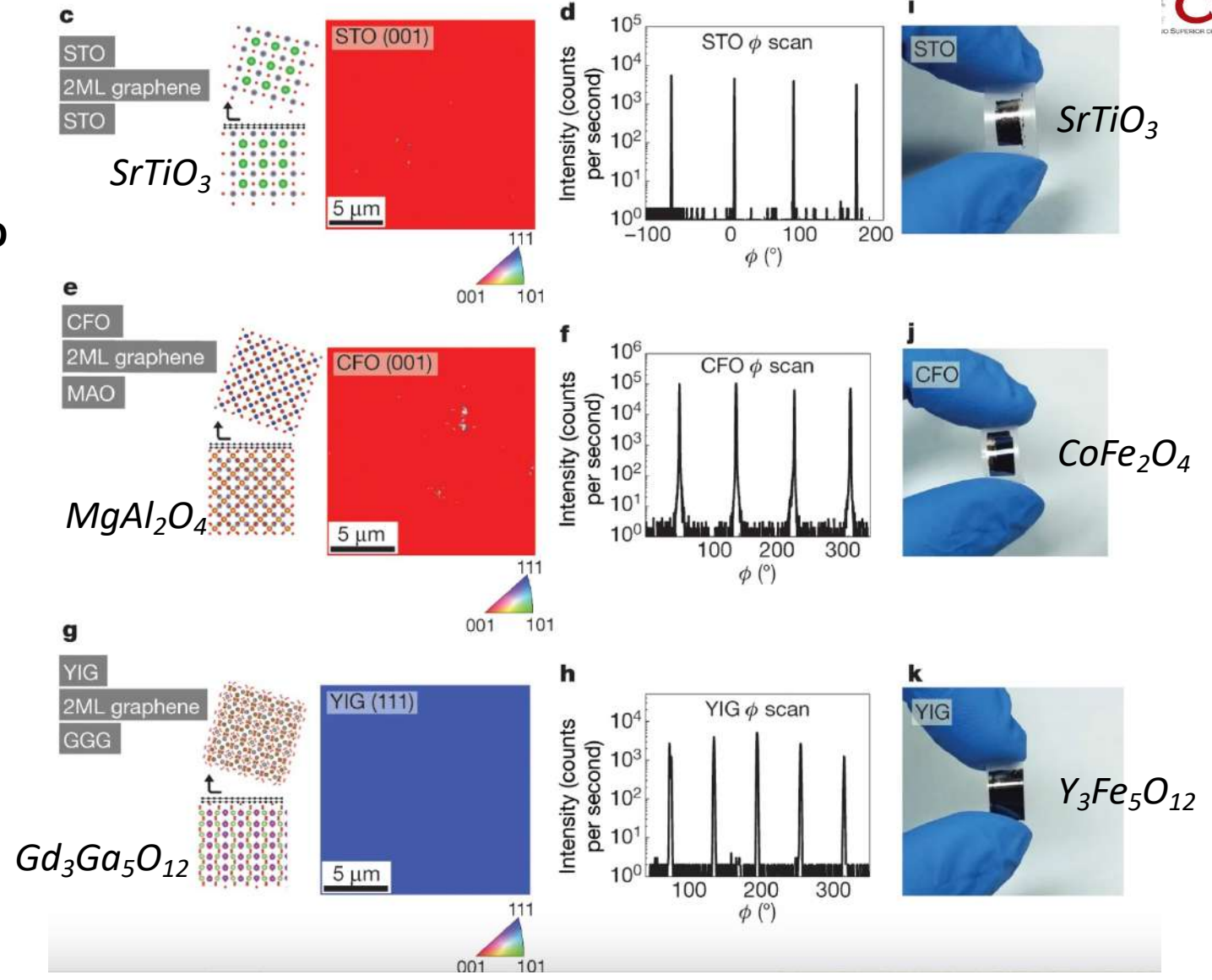
*Residues of organic/metal*  
*Film quality depends on graphene quality*  
*Still challenging large areas*

*Ex: mica or graphene--> many oxides*

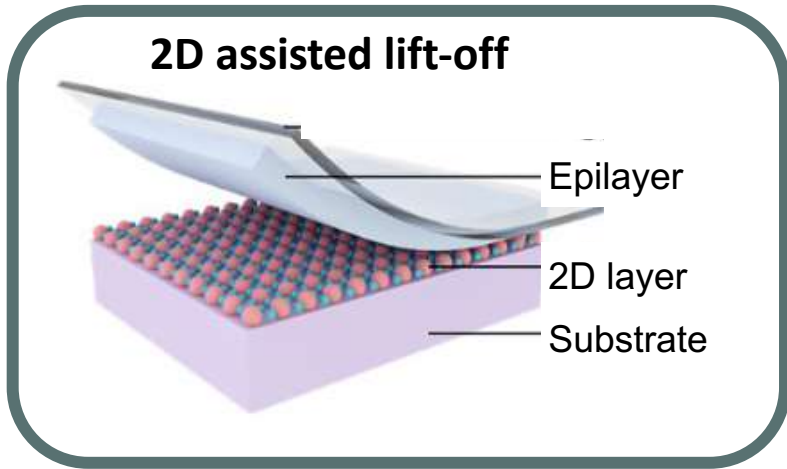
# Freestanding epitaxial complex oxides: 2D assisted lift-off. Examples



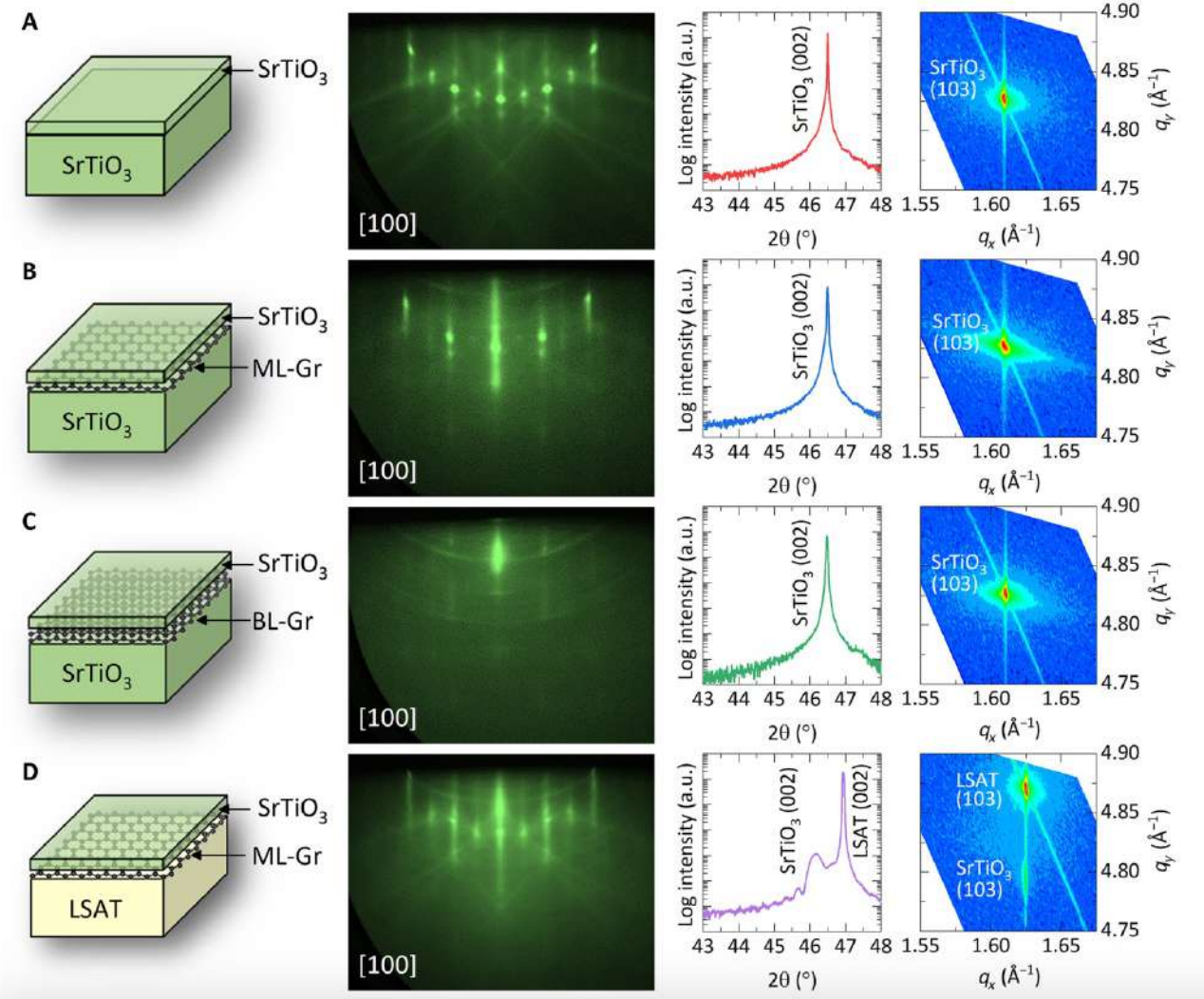
PLD







Hybrid-MBE



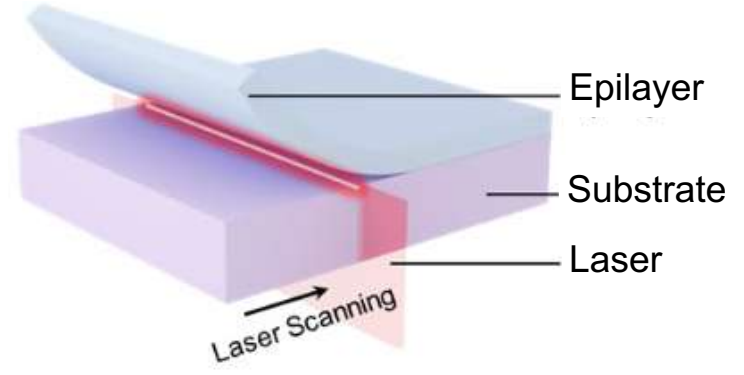
**Table 1.** Reported complex-oxides and processes for remote epitaxy.

Grown epilayer	Substrate	2D interlayer	2D synthesis method	2D transfer method	Epitaxy tool	Comment	References
STO	STO	Bilayer graphene	Graphitization of SiC	Semi-dry	PLD	Bilayer graphene and ultra-thin buffer layer are needed	[58]
CFO	MAO	Bilayer graphene	Graphitization of SiC	Semi-dry	PLD	Bilayer graphene is needed	[58]
YIG	GGG	Bilayer graphene	Graphitization of SiC	Semi-dry	PLD	Bilayer graphene is needed Post-annealing about 850 °C to improve crystal quality	[58, 94]
BTO	STO	Monolayer graphene	Graphitization of SiC/CVD on Cu foil	Semi-dry/Wet	MBE	Graphene is immediately etched if ozone is used	[58]
VO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Monolayer graphene	CVD on Cu foil	Wet	PLD	Low exfoliation area yield (5%)	[84]
LiNbO <sub>3</sub>	Al <sub>2</sub> O <sub>3</sub>	Monolayer graphene	CVD on Cu foil	Wet	PLD	There is no mention of exfoliation	[85]
STO	STO	Mono-/bilayer graphene	Direct CVD growth	—	PLD	Atomic layer etching is used Ultra-thin buffer layer is needed	[87]
VO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Bilayer graphene	CVD on Cu foil	Wet	Sputter	There is no mention of exfoliation	[86]
STO	STO	Monolayer graphene	CVD on Cu foil	Wet	MBE	Graphene is immediately etched if ozone is used regardless of buffer layer	[91]
BTO	Ge	Monolayer graphene	Direct CVD growth	—	PLD	Ge (110) substrate allows remote interaction	[88]
STO	STO	Bilayer graphene	CVD on Cu foil	Wet	MBE	Hybrid MBE is used	[92]
LSAT	STO	Monolayer graphene	CVD on Cu foil	Wet	MBE	Hybrid MBE is used	[92]

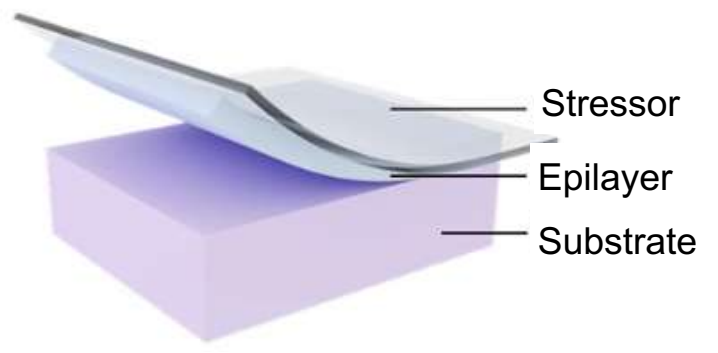
# Freestanding epitaxial complex oxides: How to detach them from the substrate?

## PHYSICAL TECHNIQUES

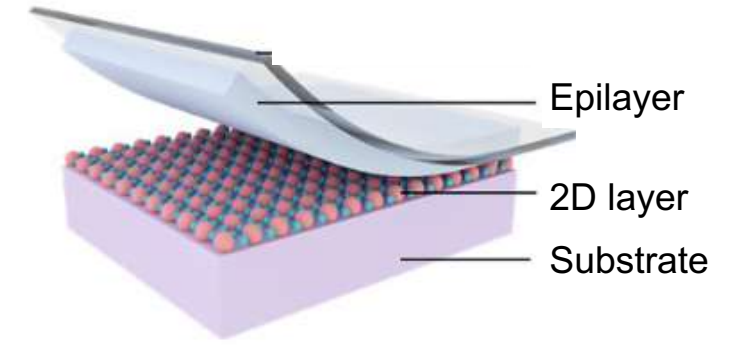
### Laser lift-off



### Mechanical lift-off



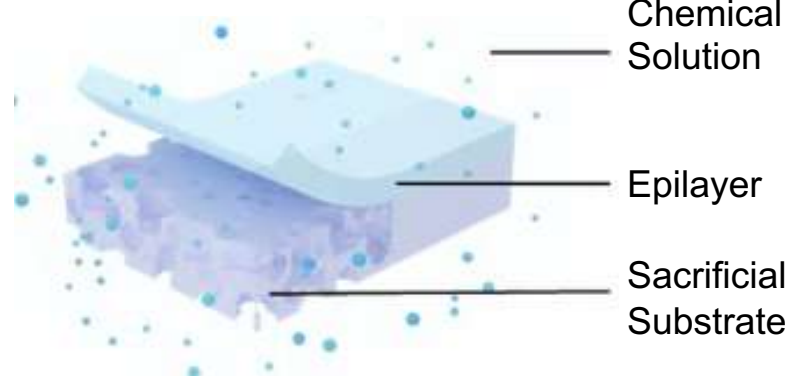
### 2D assisted lift-off



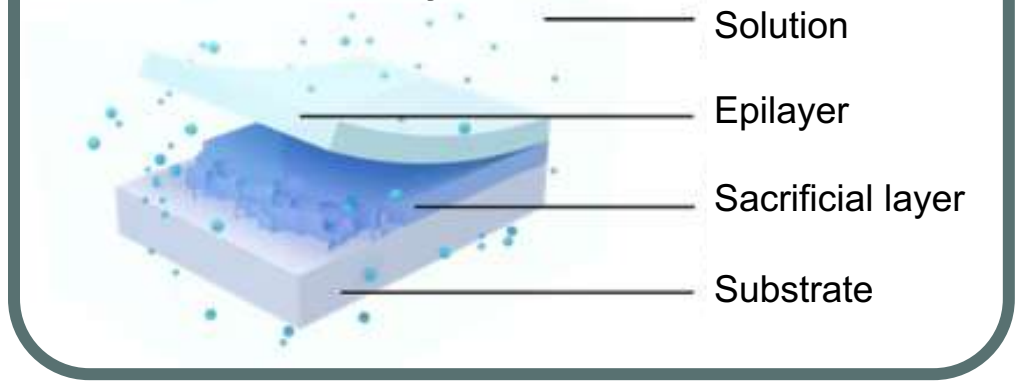
HS Kum *et al.*, *Nature* **578**, 75–81, 2020

## CHEMICAL TECHNIQUES

### Substrate etching



### Sacrificial layer



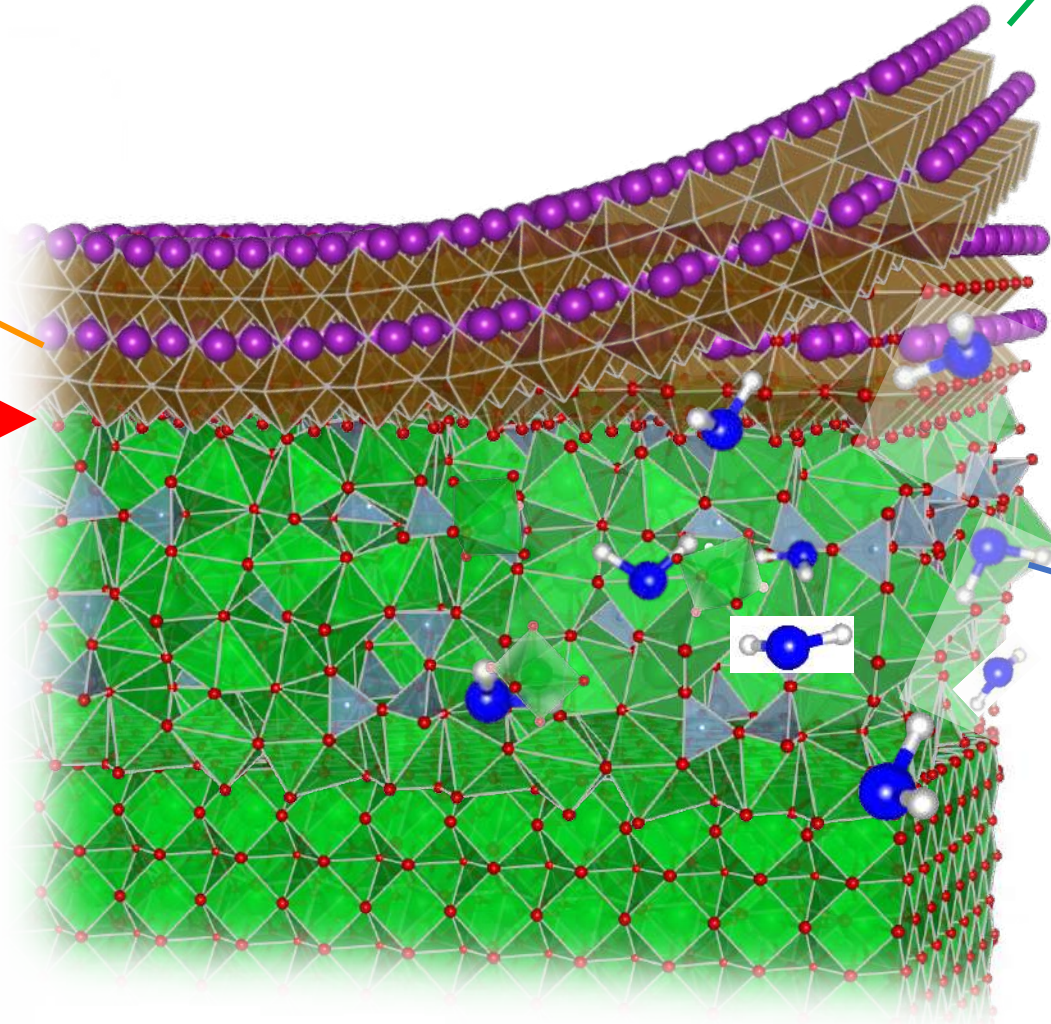
# Sacrificial

Oxide transfer  
Towards device  
fabrication

Epitaxy, stoichiometry  
Continuous film (crack and wrinkles free)  
Interaction/adhesion with arbitrary  
substrates (polymer, semiconductors)

Functional  
oxide  
 $\text{CoFe}_2\text{O}_4$   
 $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$   
 $\text{BiFeO}_3$   
....

Epitaxy  
Interdiffusion  
Sharp interface  
Thickness  
Functionality



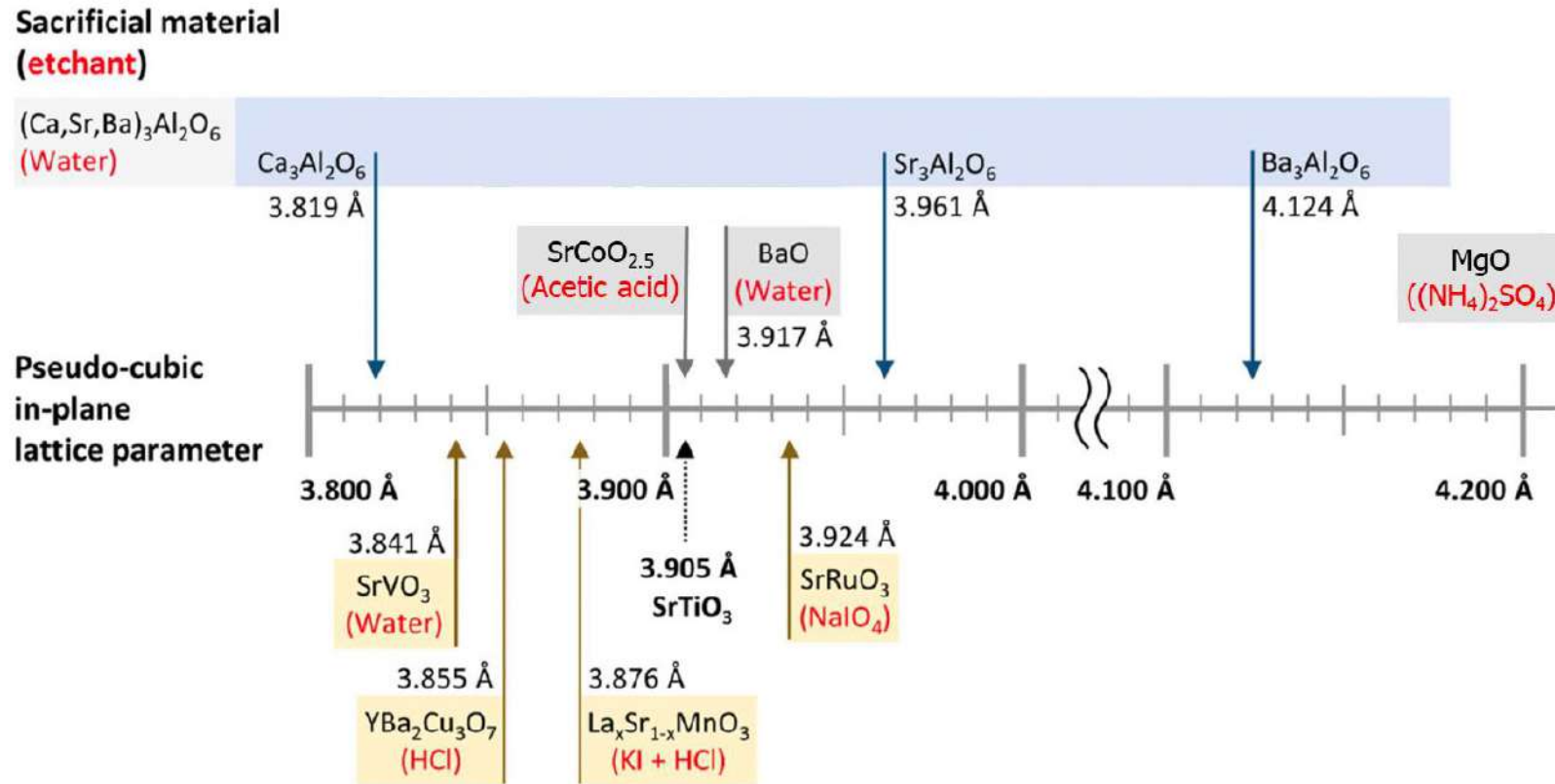
Sacrificial Layer

Selective etching  
Epitaxy  
Smooth Surface  
Stability  
Thickness

**Table 1.1:** Sacrificial layer compositions for the preparation of freestanding epitaxial complex oxide films.

Sacrificial Layer	Substrate	Etching solution	Released epitaxial oxide
$\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$	$\text{SrTiO}_3$ (001)	KI + HCl	$\text{BaTiO}_3$ , <sup>31,51</sup> PZT, $\text{CaTiO}_3/\text{SrTiO}_3$ superlattice <sup>29</sup>
$\text{SrRuO}_3$	$\text{SrTiO}_3$ (001)	$\text{NaIO}_{4(\text{aq})}$	LSMO <sup>52</sup>
$\text{YBa}_2\text{Cu}_3\text{O}_7$	$\text{SrTiO}_3$ (001)	HCl	LSMO <sup>53</sup>
$\text{SrVO}_3$	$\text{SrTiO}_3$ (001)	50 °C Water	$\text{SrTiO}_3$ <sup>54</sup>
$\text{SrCoO}_{2.5}$	$\text{SrTiO}_3$ (001), (110), (111)	Weak acids	$\text{SrRuO}_3$ <sup>55</sup>
MgO	$\text{SrTiO}_3$ (001)	$(\text{NH}_4)_2\text{SO}_{4(\text{aq})}$	$\text{CoFe}_2\text{O}_4$ <sup>56</sup>
BaO	$\text{SrTiO}_3$ (001)	RT Water	$\text{BaTiO}_3$ , $\text{SrTiO}_3$ <sup>57</sup>
$(\text{Ca,Sr,Ba})_3\text{Al}_2\text{O}_6$	$\text{SrTiO}_3$ (001), (110), (111), $\text{LaAlO}_3$ (001)	RT Water	See 1.1.3 $\text{Sr}_3\text{Al}_2\text{O}_6$ (SAO) sacrificial layer

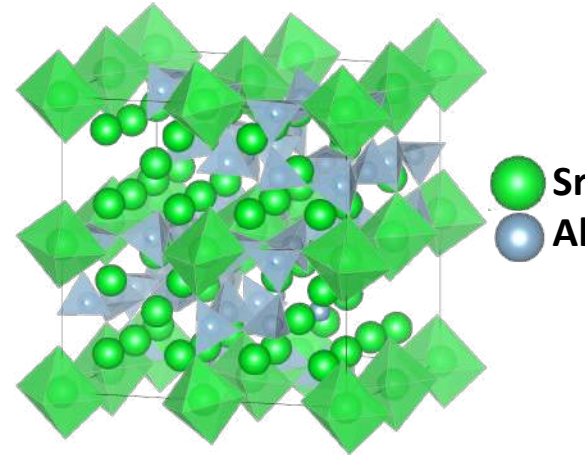
# Sacrificial compatibility with epitaxial oxides



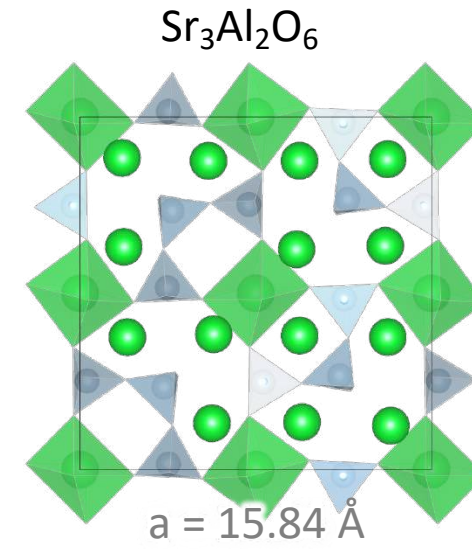
**Figure 1.5: Compatible sacrificial layers with epitaxial complex oxides.** Pseudo-cubic in-plane lattice parameters of the different sacrificial layer compositions included in Table 1.1. The yellow frames correspond to perovskites, the blue one corresponds to the  $(\text{Ca,Sr,Ba})_3\text{Al}_2\text{O}_6$  family and the grey ones to other alternative structures.  $\text{SrTiO}_3$ , as typical single-crystal substrate for the growth of epitaxial complex oxides, is depicted as reference. *Figure adapted from Ref.<sup>58</sup>.*

# Sr<sub>3</sub>Al<sub>2</sub>O<sub>6</sub> (SAO) as sacrificial layer

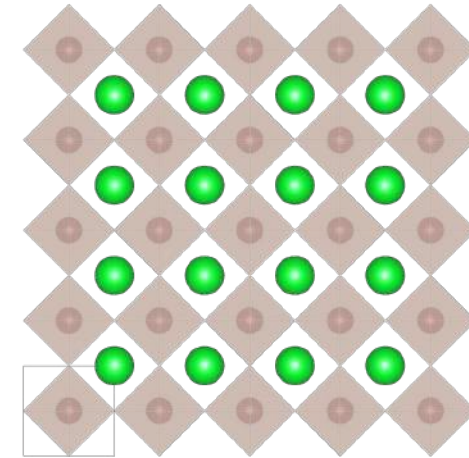
- ✓ Similar cell parameters to SrTiO<sub>3</sub>, La<sub>0.7</sub>Sr<sub>0.3</sub>MnO<sub>3</sub>, BiFeO<sub>3</sub>,... (epitaxial growth)
- ✓ Dissolved in H<sub>2</sub>O (non-toxic, low-cost, **soft**)
- ✓ High versatility



Pseudo-perovskite structure

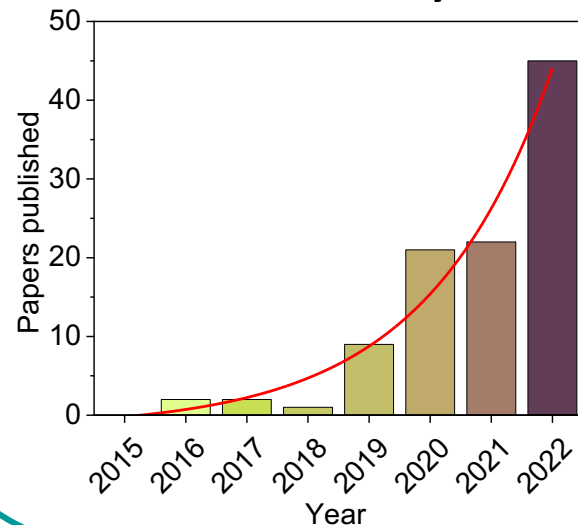


D Lu et al., *Nat. Mater.* **15**, 1255–1260, 2016

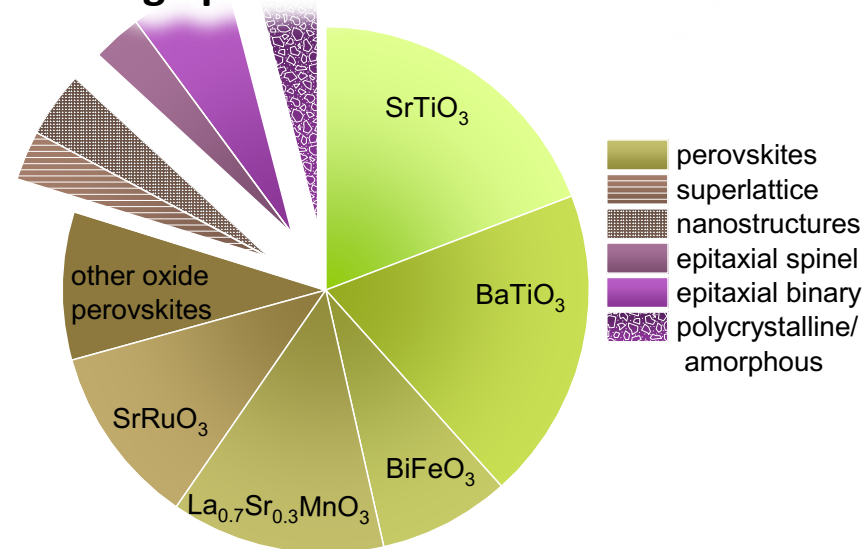


## STATE OF THE ART (Last update: 31<sup>st</sup> Dec 2022)

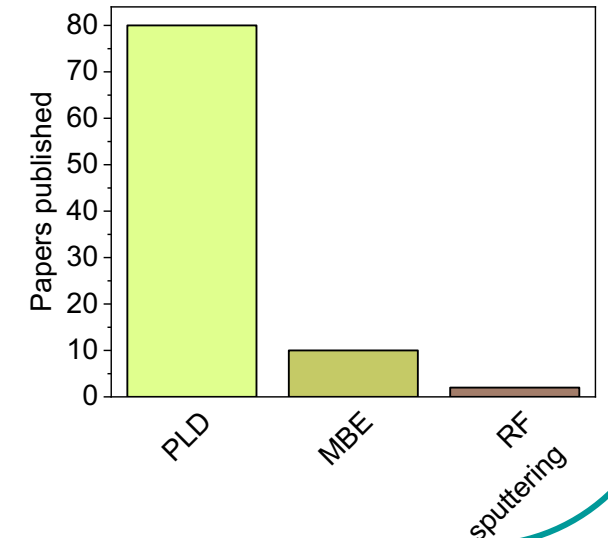
### Papers published using SAO as sacrificial layer



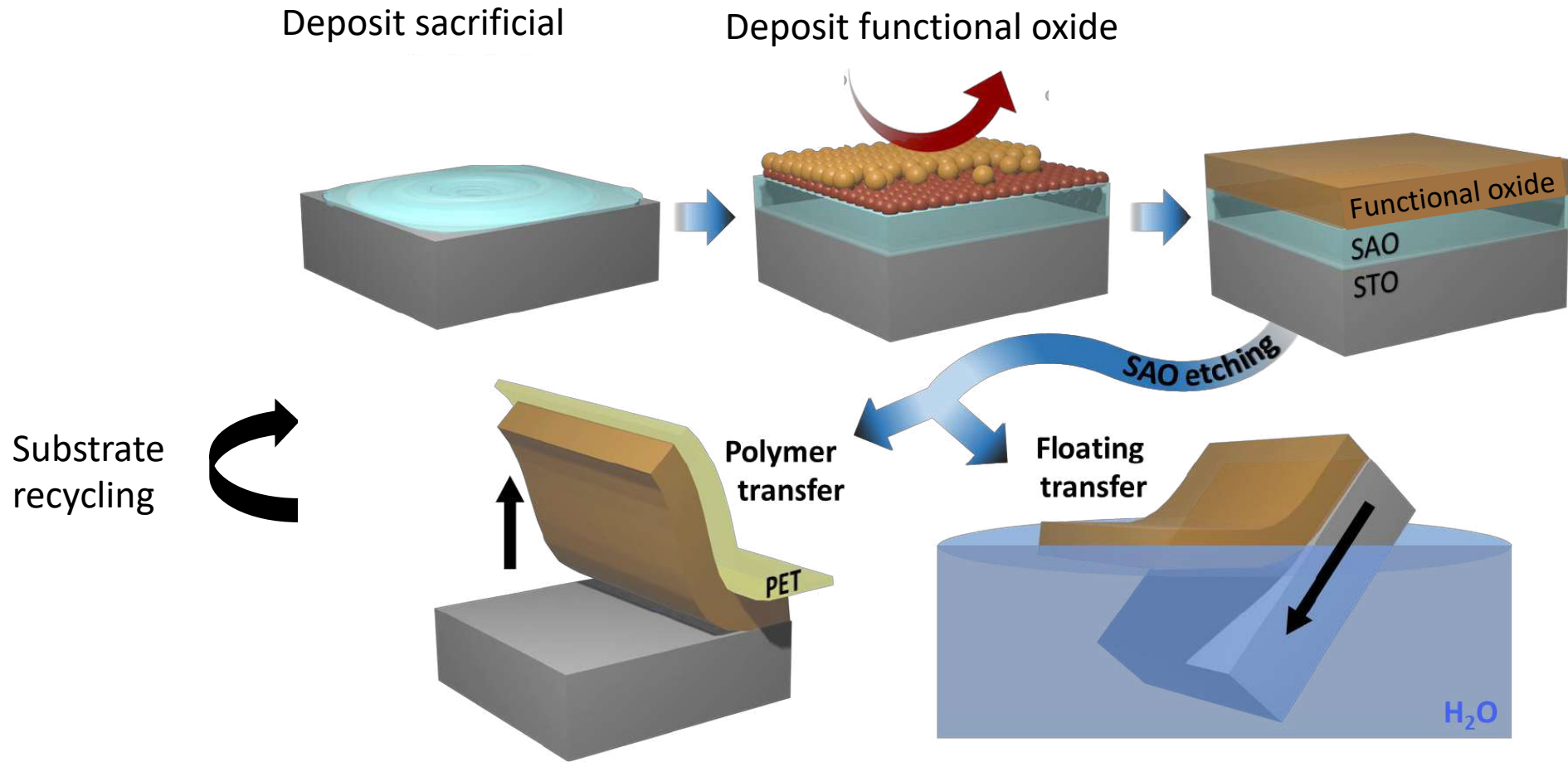
### Freestanding epitaxial oxide structures



### High-vacuum deposition techniques

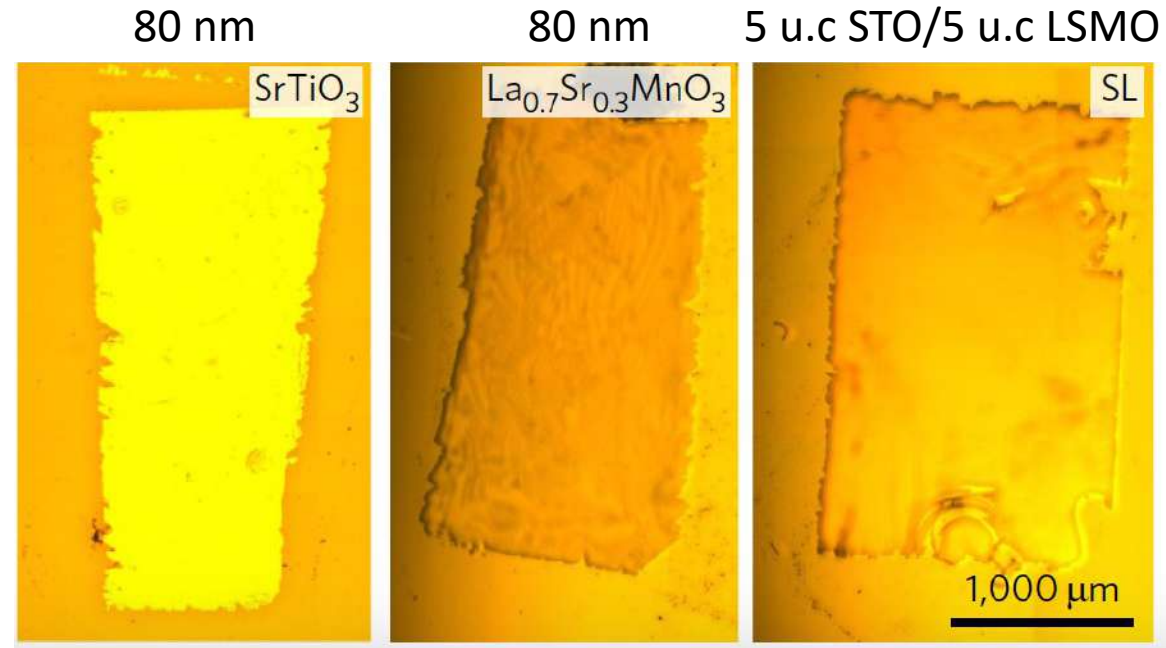


# Schematic of the procedure to obtain functional oxide membranes...

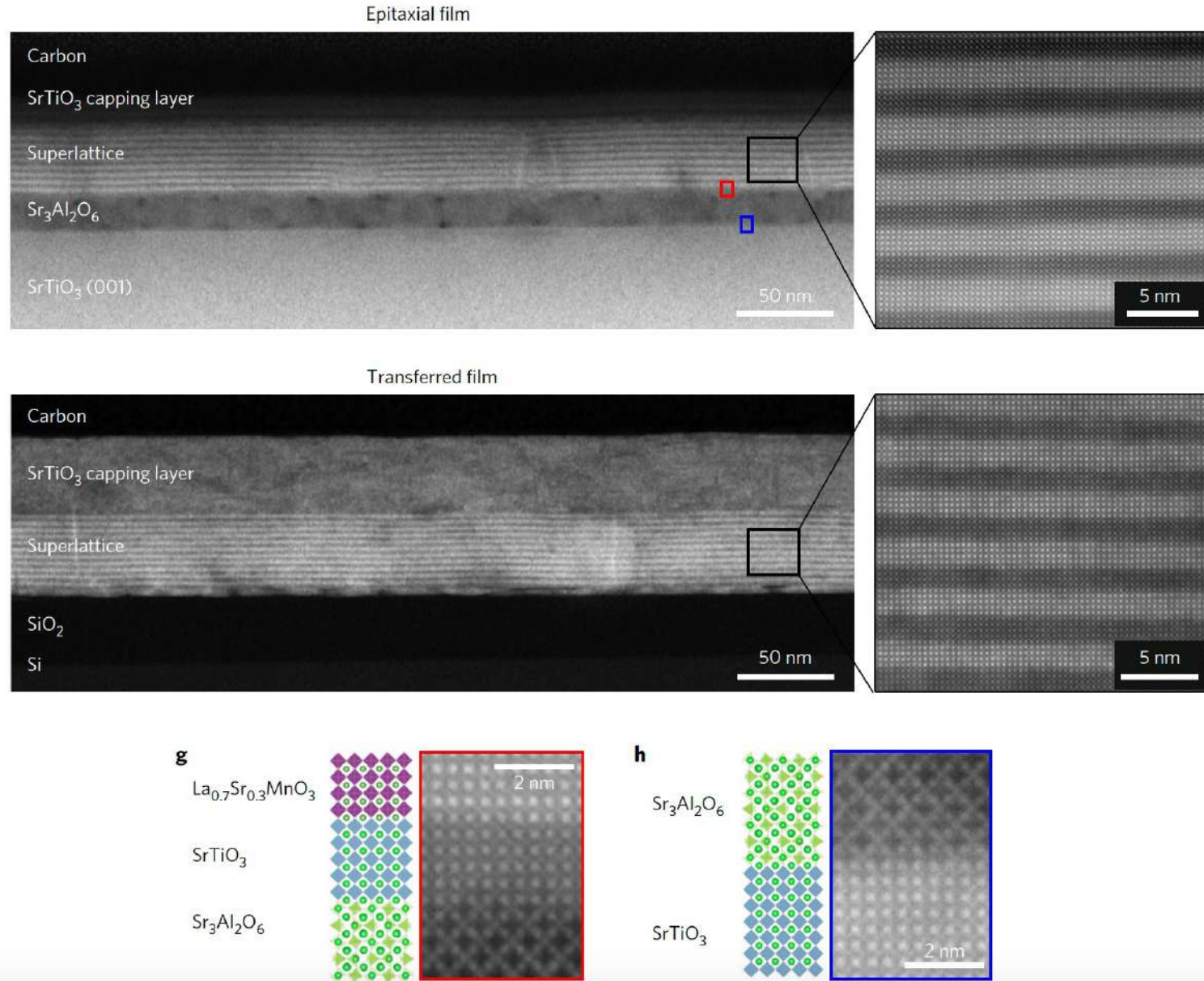




# Synthesis of freestanding single-crystal perovskite films and heterostructures

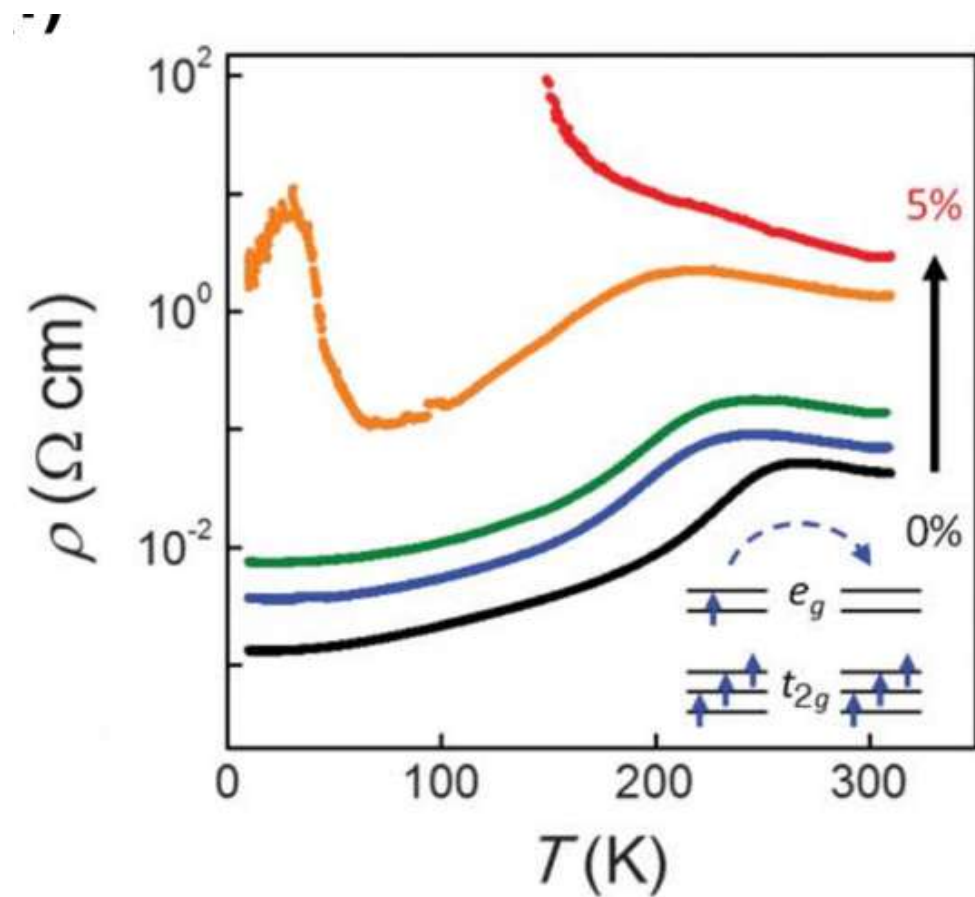
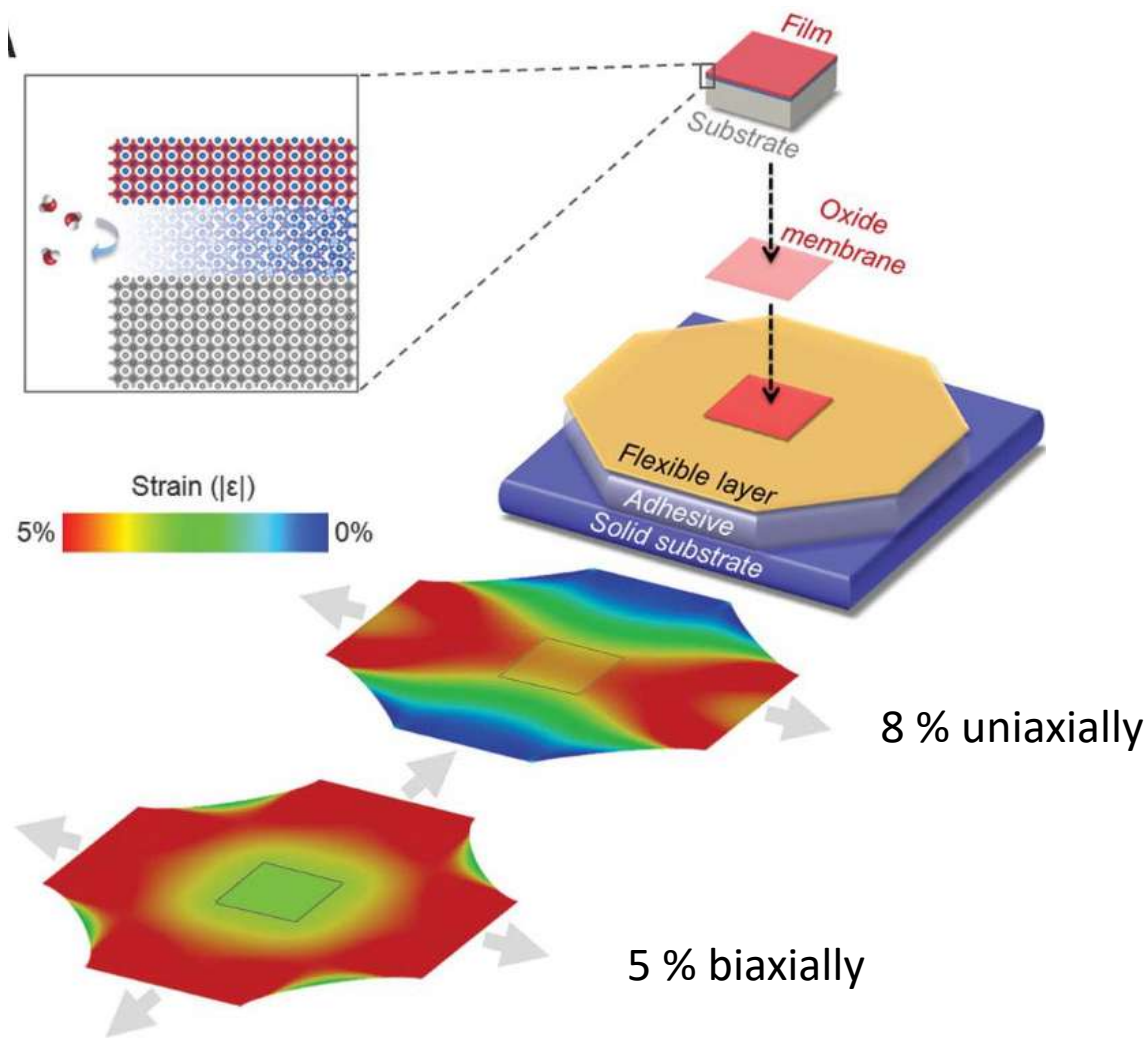


# Synthesis of freestanding single-crystal perovskite films and heterostructures



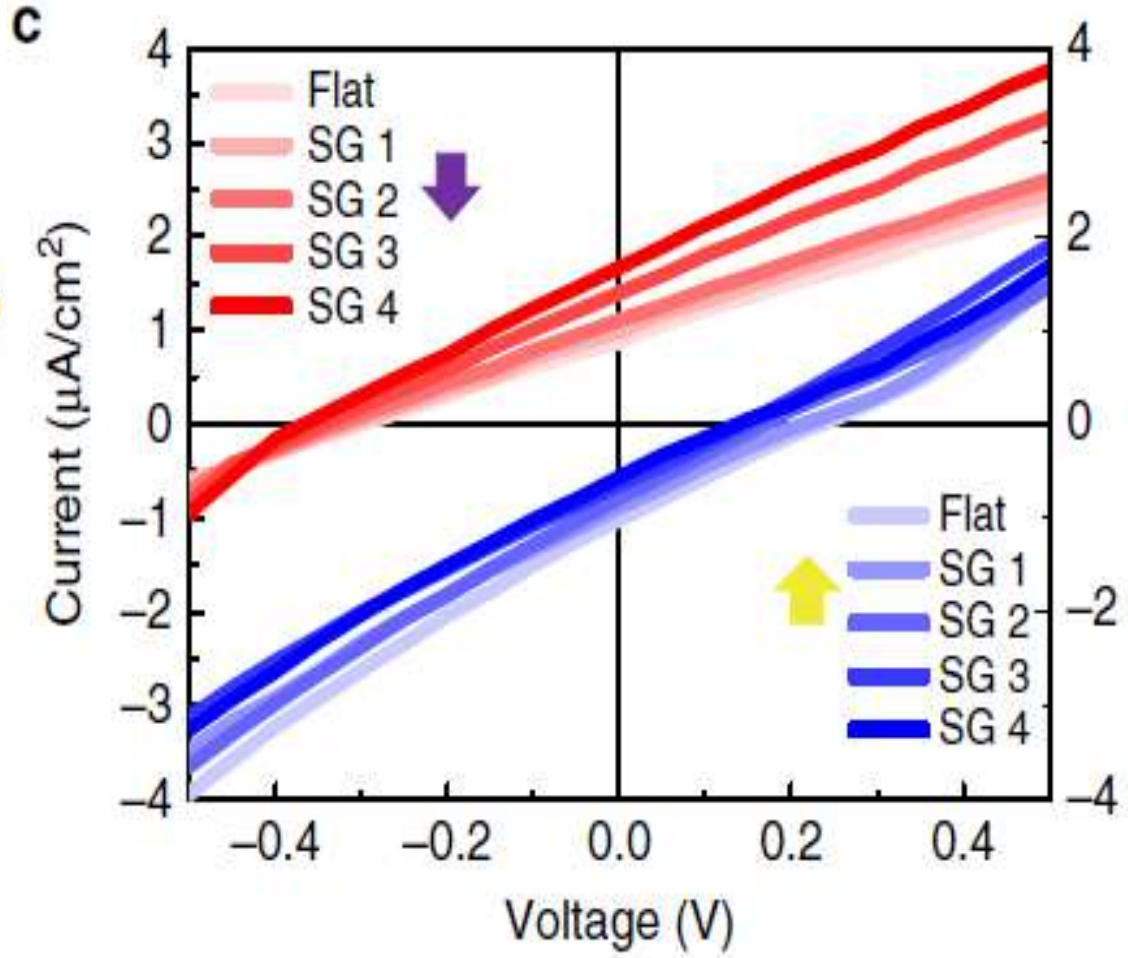
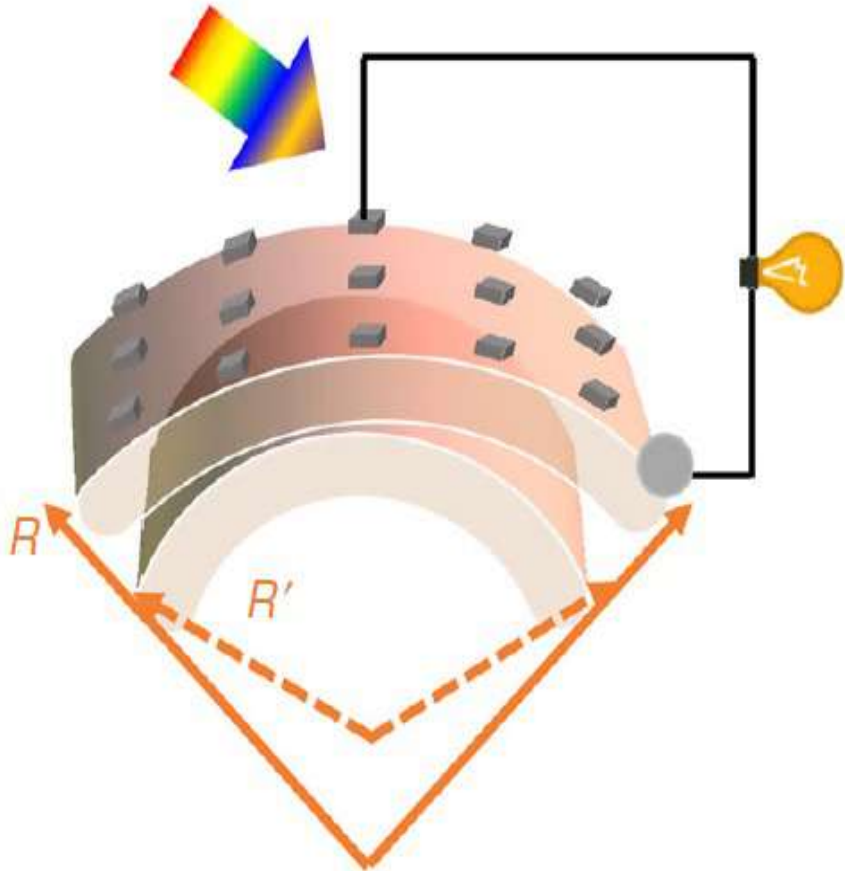
# Novel functionalities in freestanding oxides

Extreme tensile strain states  $\text{La}_{0.7}\text{Ca}_{0.3}\text{MnO}_3$  :metallic vs insulating



# Novel functionalities in freestanding oxides

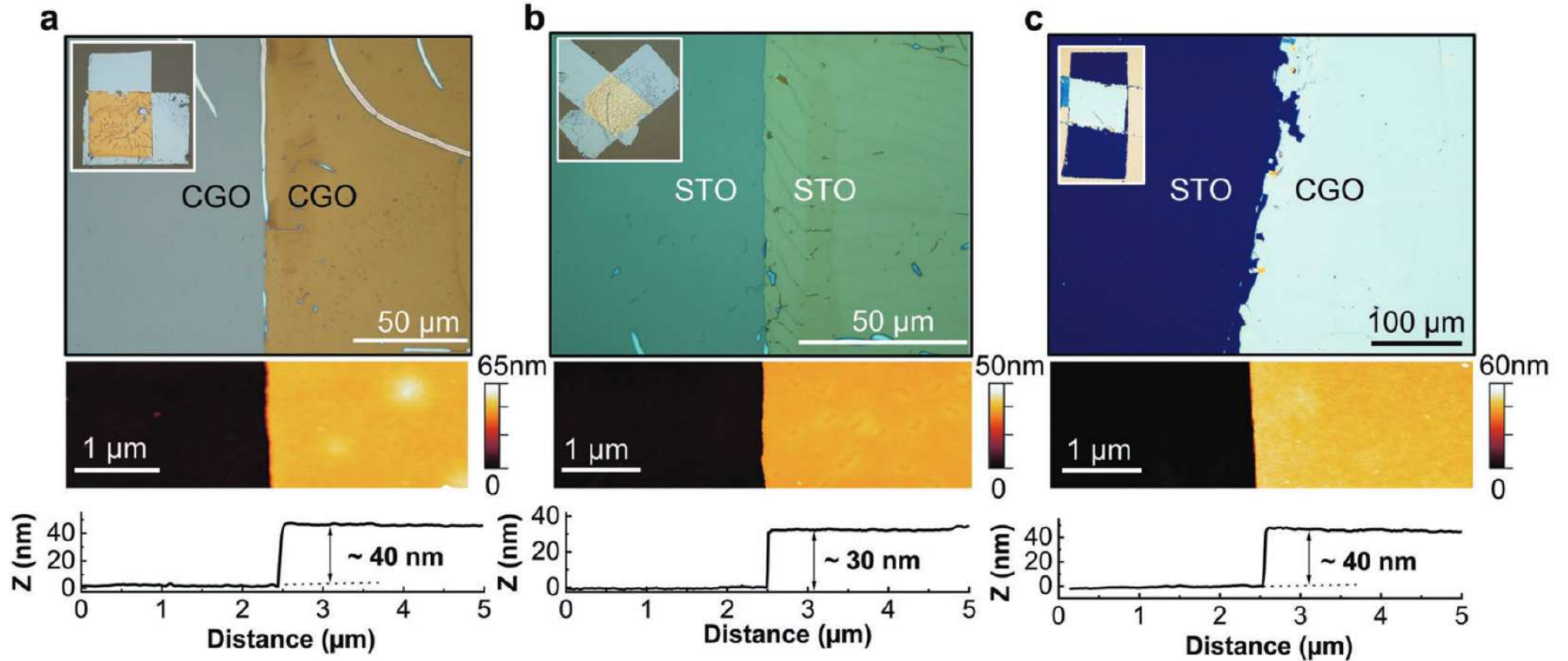
- Flexoelectricity by BFO bending: controllable photoconductance



non uniform lattice distortion--> strain gradient: multilevel photoconductance

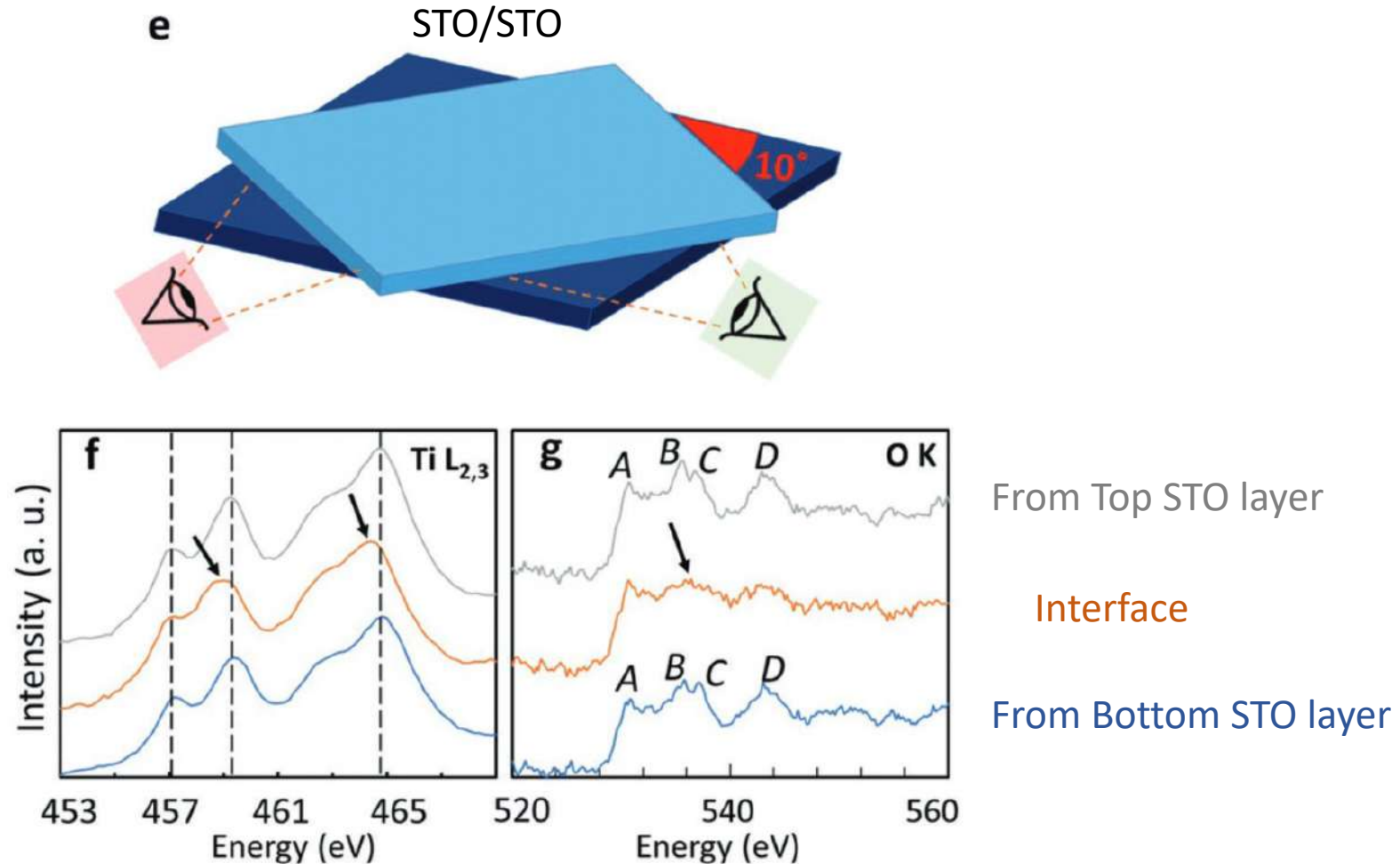
# Novel functionalities in freestanding oxides

- Artificial interfaces+ twisting angle



# Novel functionalities in freestanding oxides

- Artificial interfaces+ twisting angle

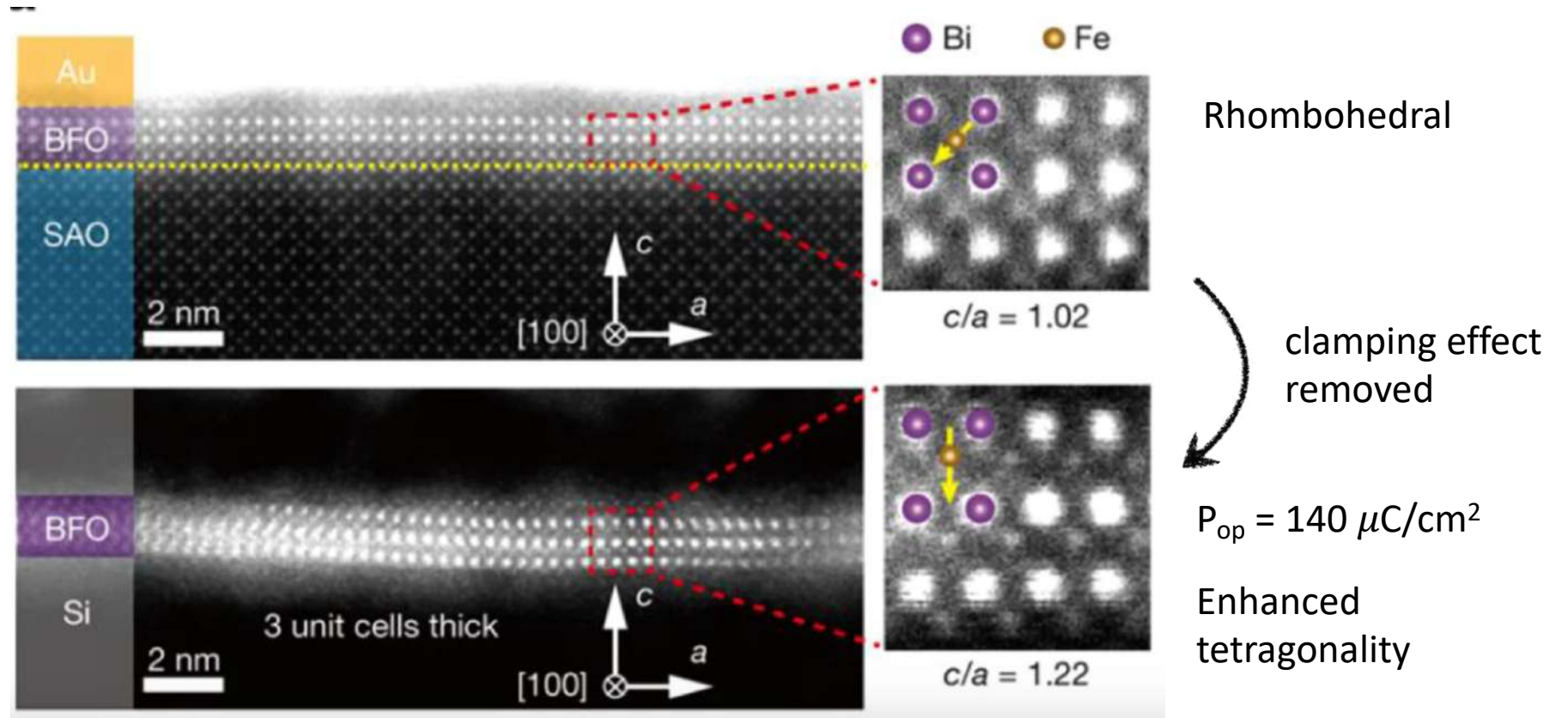


“twisting” yields anisotropic oxygen ion diffusion @ STO/STO stack with twist angle  $\theta$ . Otherwise isotropic in bulk (001)-STO

Preference for oxygen vacancies--> strategy to manipulate diffusion properties of  $V_o$  in perovskite oxides

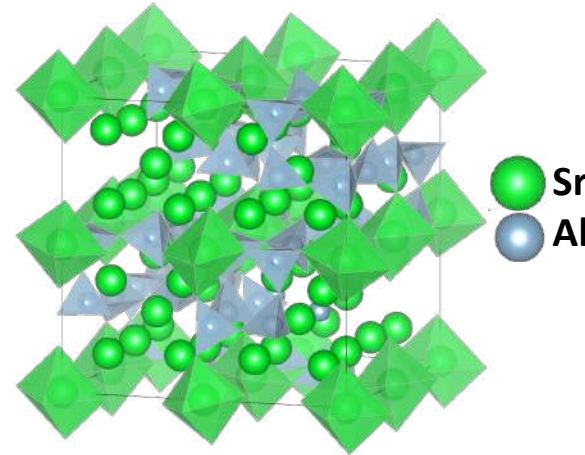
# Novel functionalities in freestanding oxides

- Approaching the monolayer limit

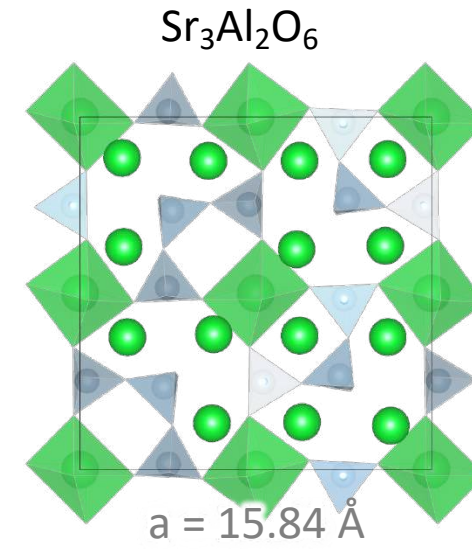


# Sr<sub>3</sub>Al<sub>2</sub>O<sub>6</sub> (SAO) as sacrificial layer

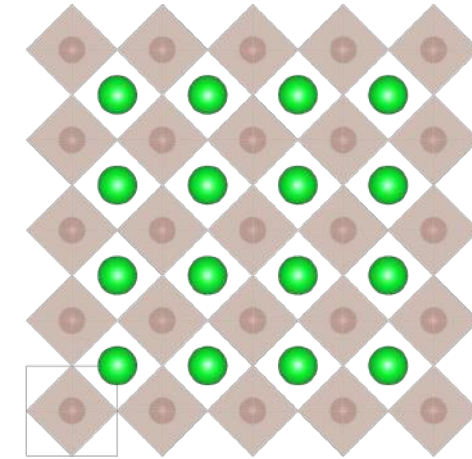
- ✓ Similar cell parameters to SrTiO<sub>3</sub>, La<sub>0.7</sub>Sr<sub>0.3</sub>MnO<sub>3</sub>, BiFeO<sub>3</sub>,... (epitaxial growth)
- ✓ Dissolved in H<sub>2</sub>O (non-toxic, low-cost, **soft**)
- ✓ High versatility



Pseudo-perovskite structure

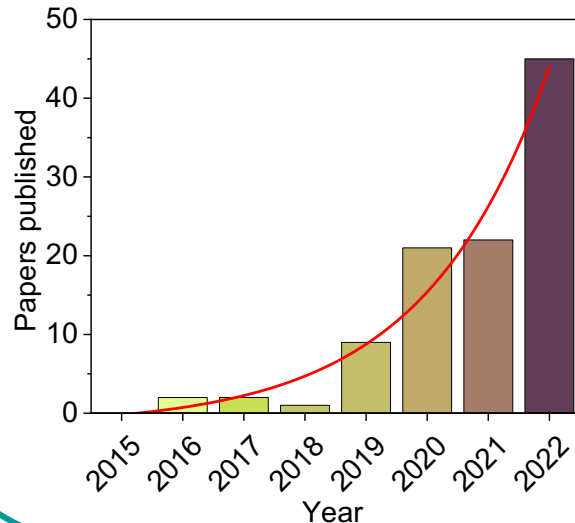


D Lu et al., *Nat. Mater.* **15**, 1255–1260, 2016

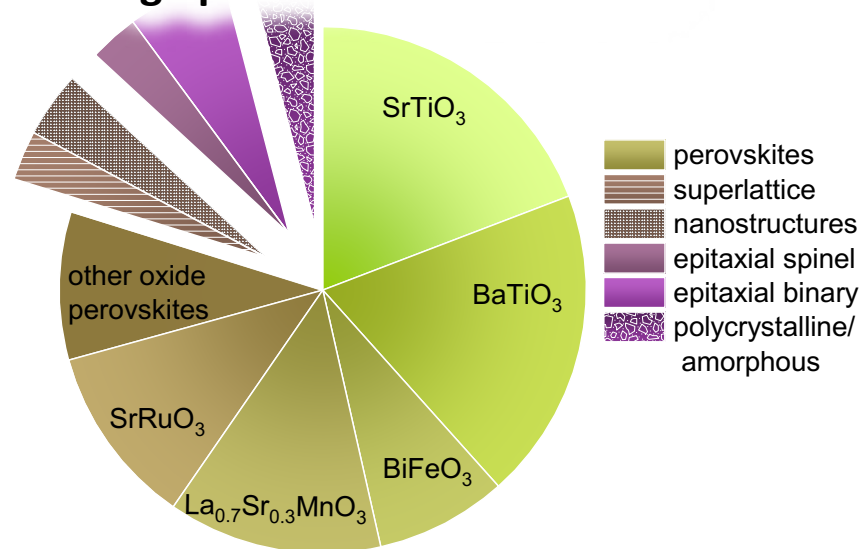


## STATE OF THE ART (Last update: 31<sup>st</sup> Dec 2022)

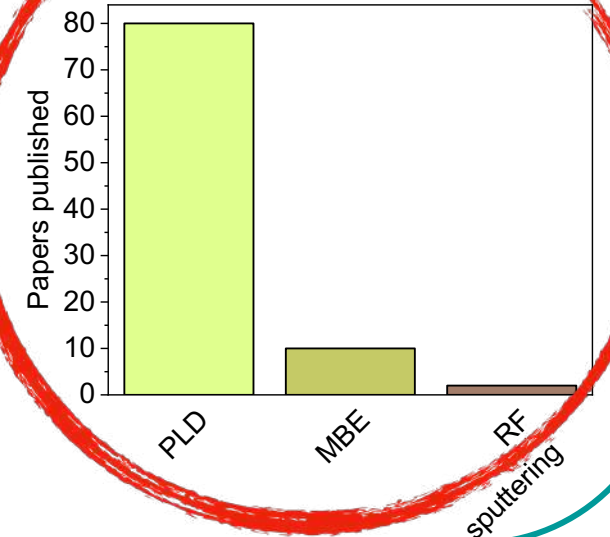
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### Freestanding epitaxial oxide structures



### High-vacuum deposition techniques



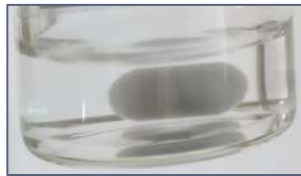
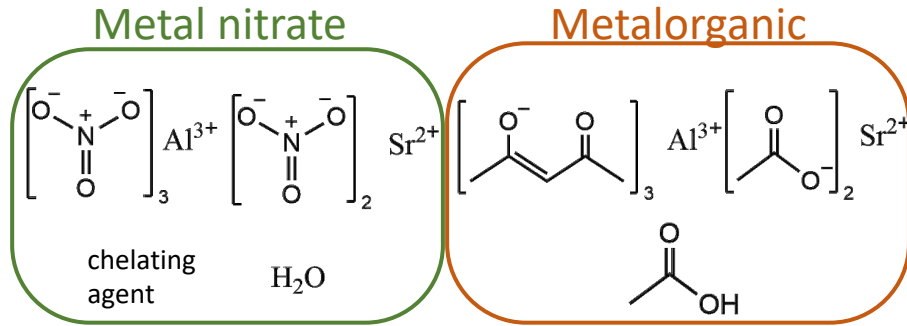


# Sr<sub>3</sub>Al<sub>2</sub>O<sub>6</sub> by chemical methods

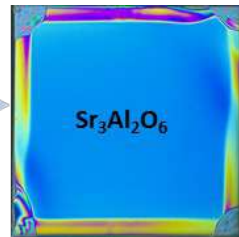
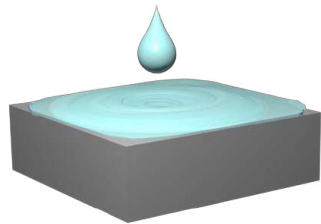


Pol Sallés

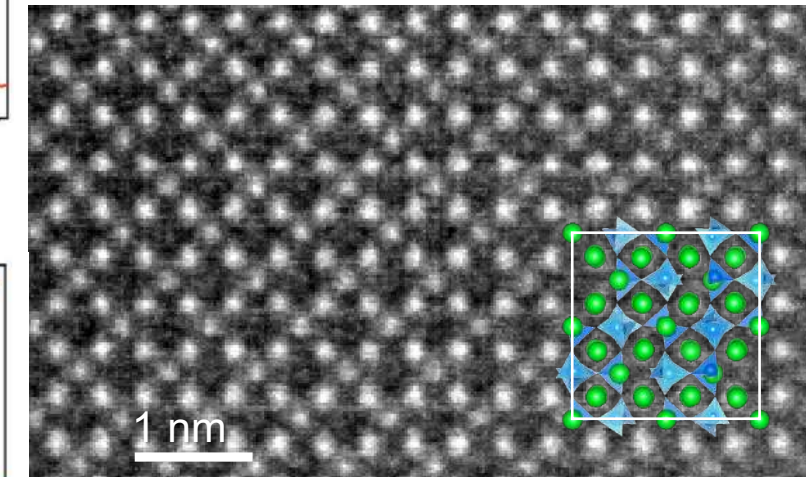
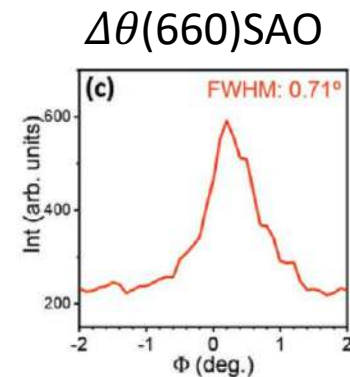
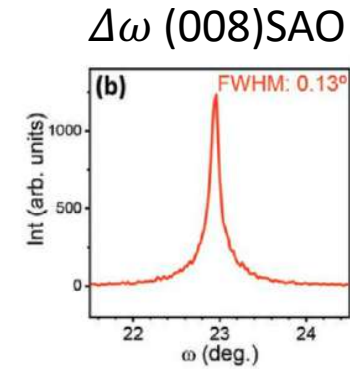
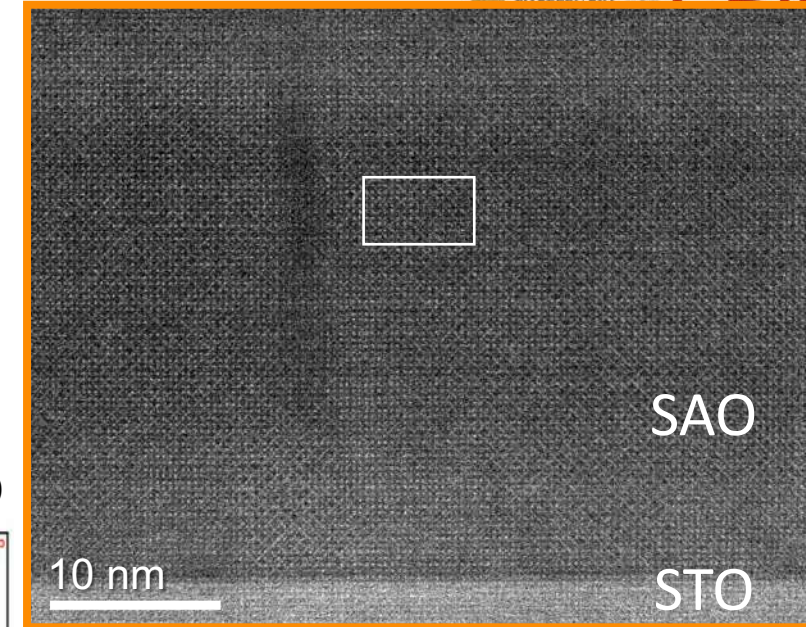
Epitaxial growth - HR-TEM cross section



Sr<sub>3</sub>Al<sub>2</sub>O<sub>6</sub> by CSD

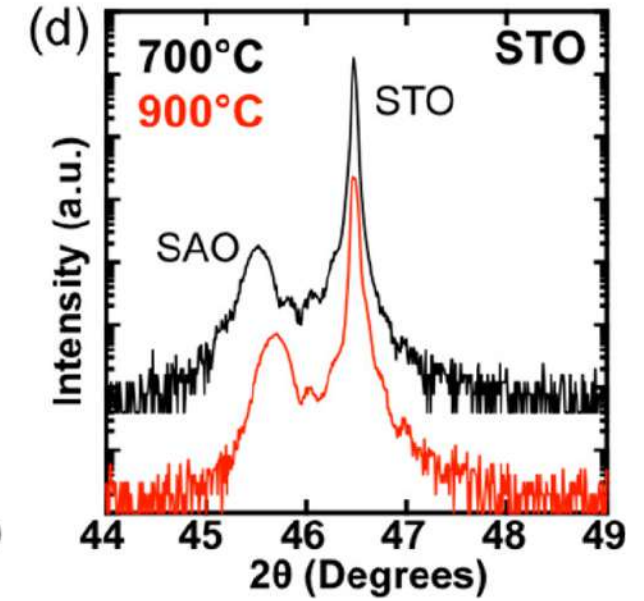
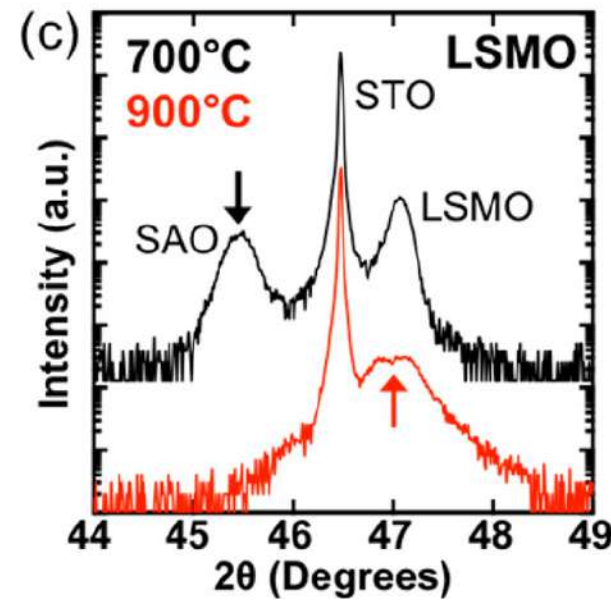
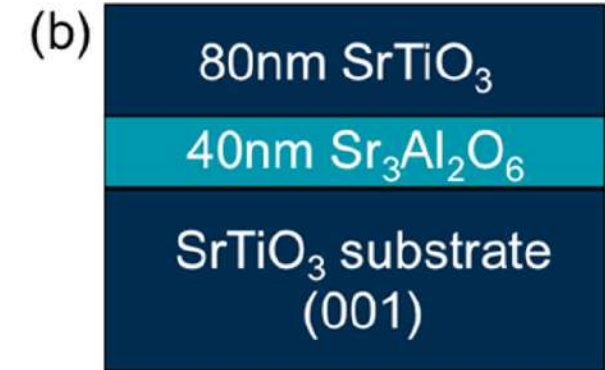
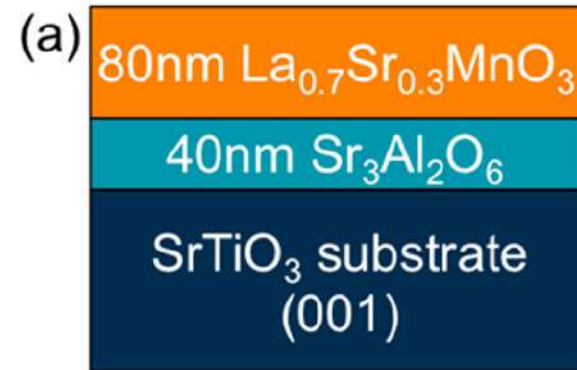
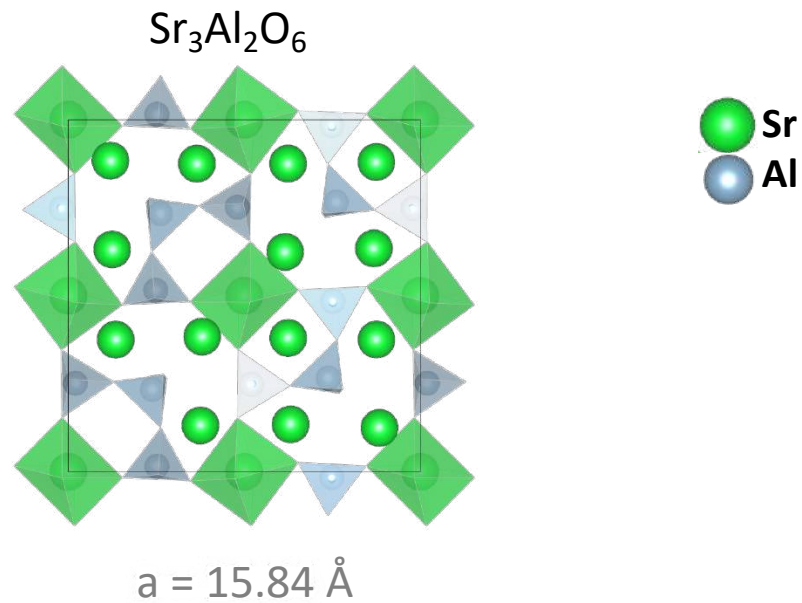


- Metal nitrate vs. metalorganic precursors
- Solvents (H<sub>2</sub>O, acetic acid)
- Chelating agent



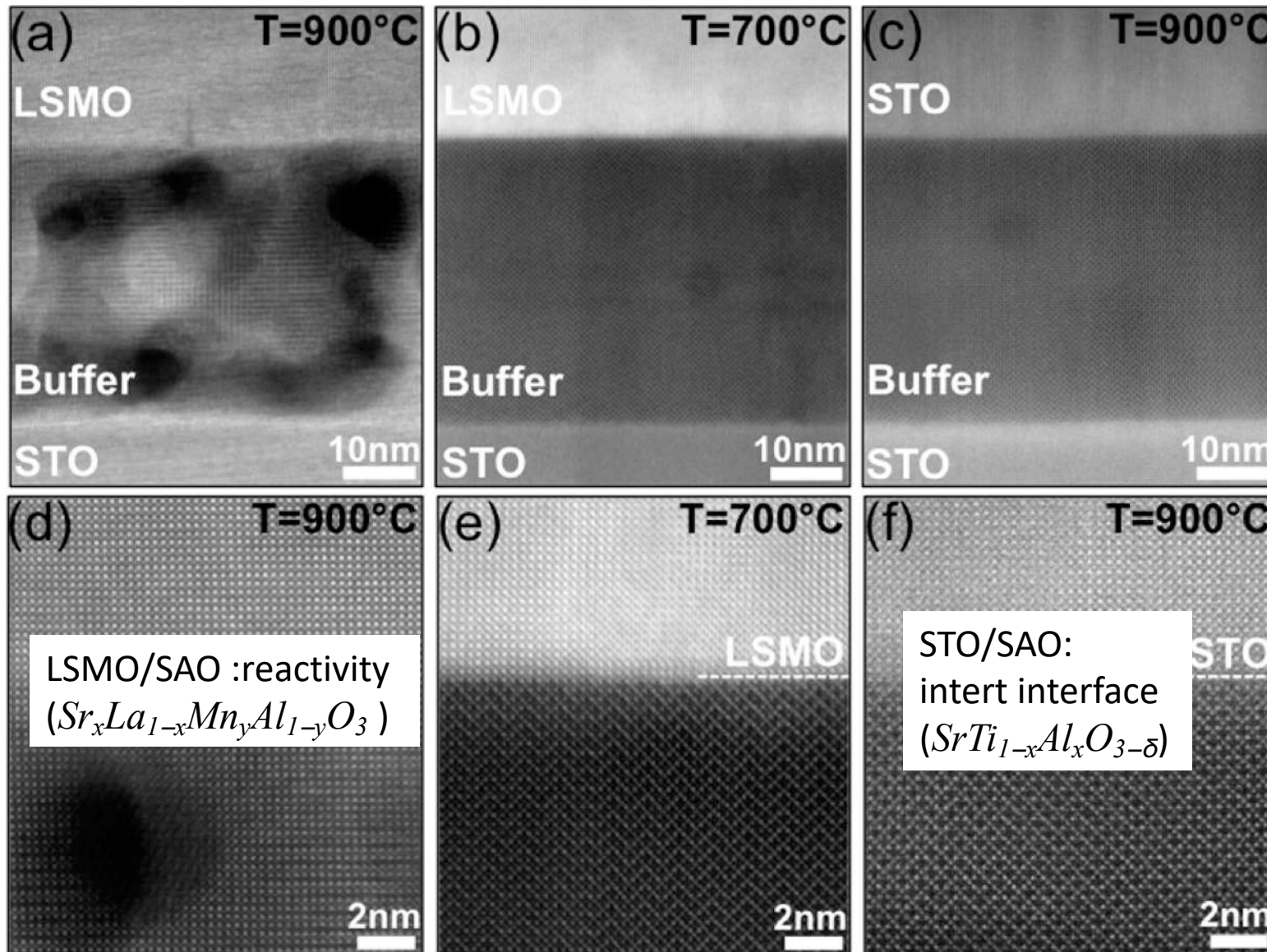
- Why free-standing oxides
- Preparation of free-standing oxides
- Challenges in the preparation
  - Oxide/Sacrificial interface
  - Exfoliation
- Outlook/ perspective

# Challenges : Oxide/ SAO interface



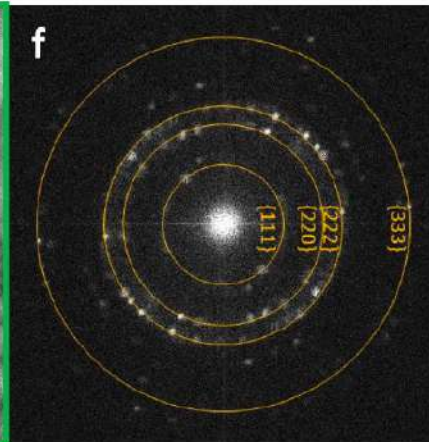
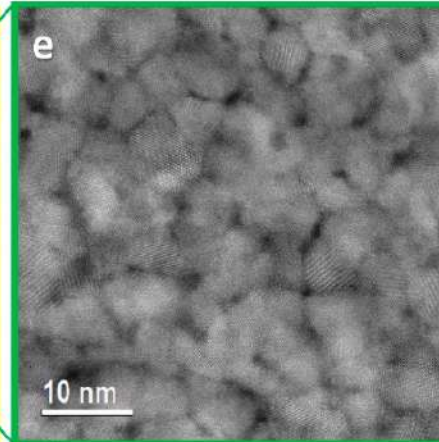
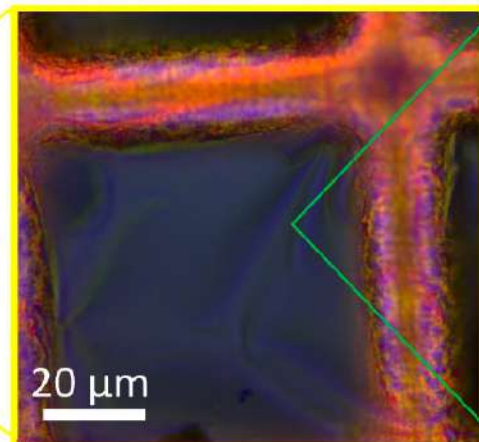
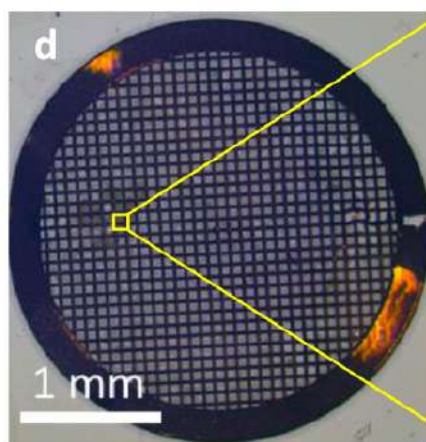
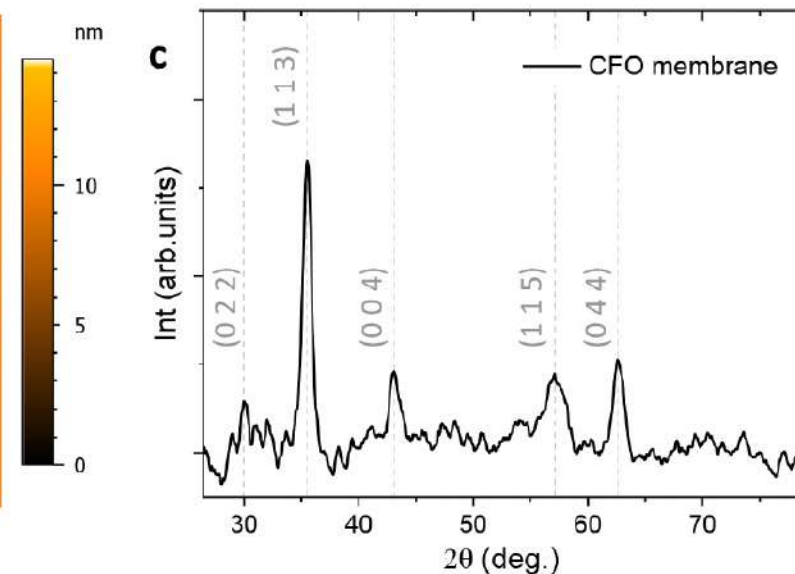
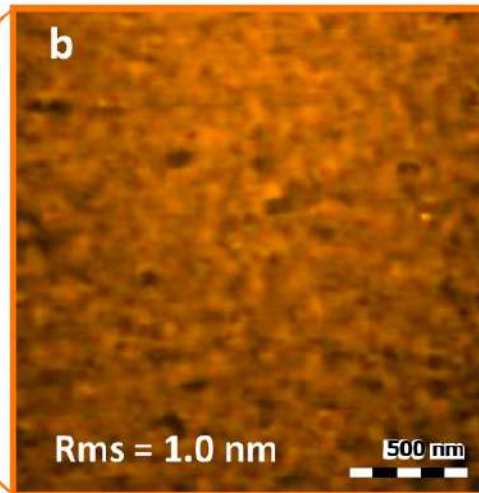
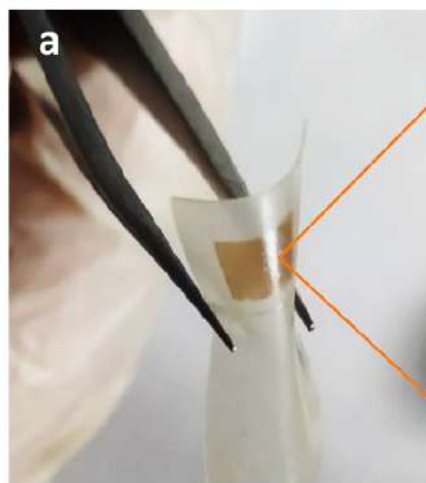
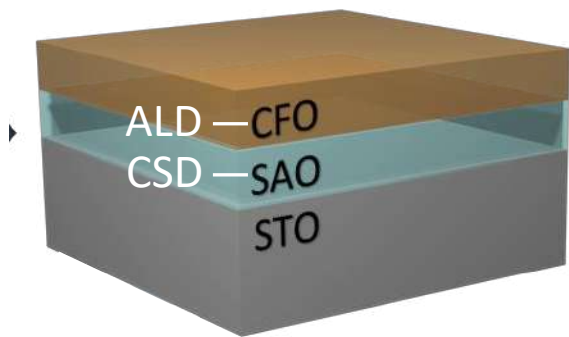
Temperature + composition

# Challenges : Oxide/ SAO interface

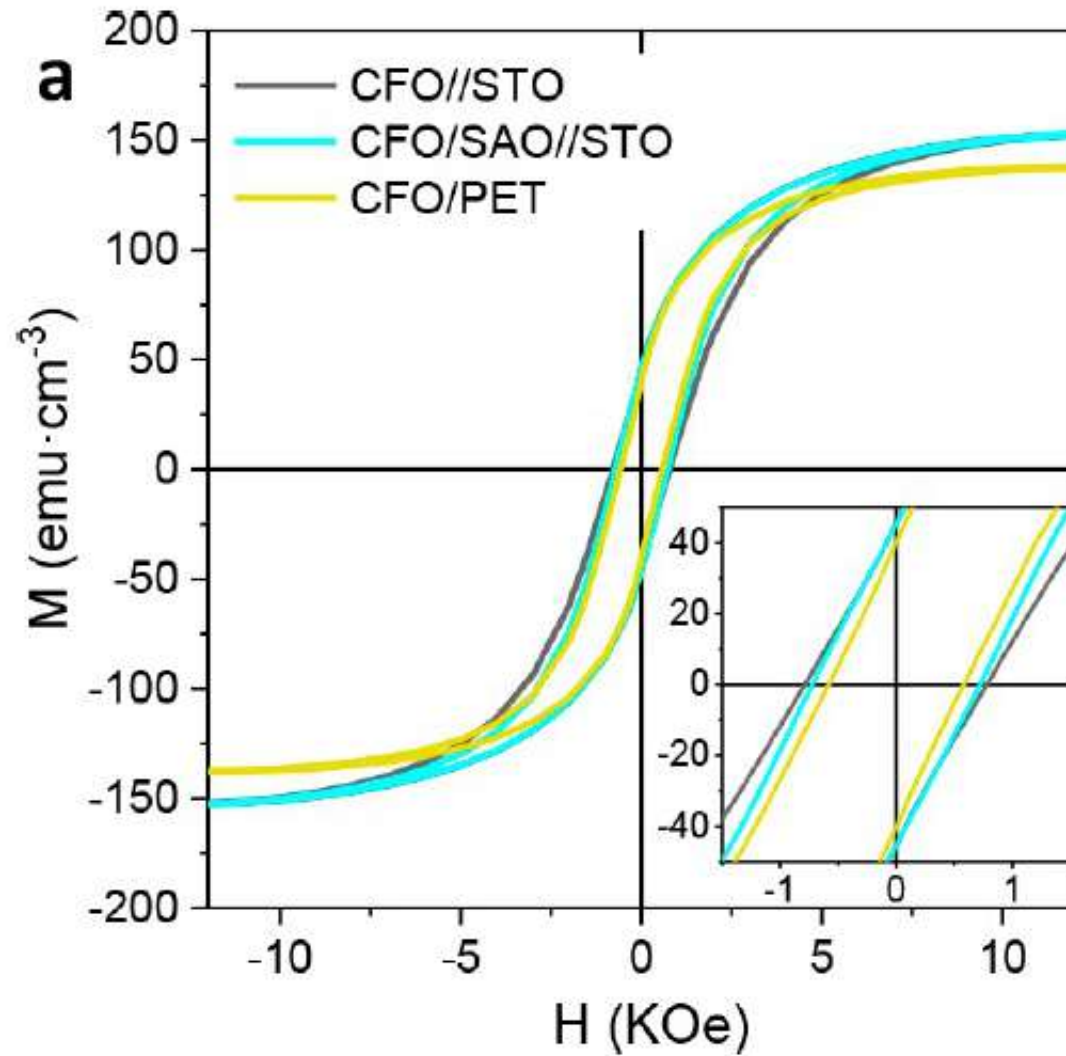


# Challenges : Oxide/SAO interface

## Bendable Polycrystalline and Magnetic $\text{CoFe}_2\text{O}_4$ Membranes : all chemical

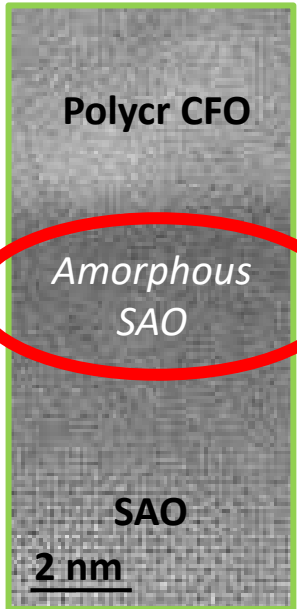


# Bendable Polycrystalline and Magnetic $\text{CoFe}_2\text{O}_4$ Membranes

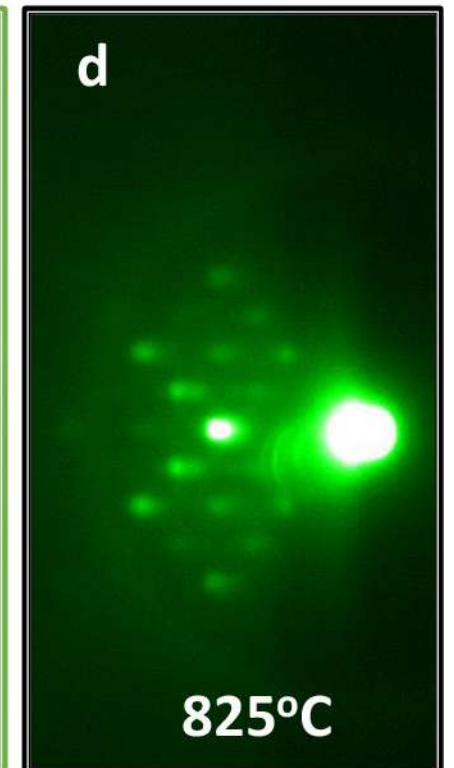
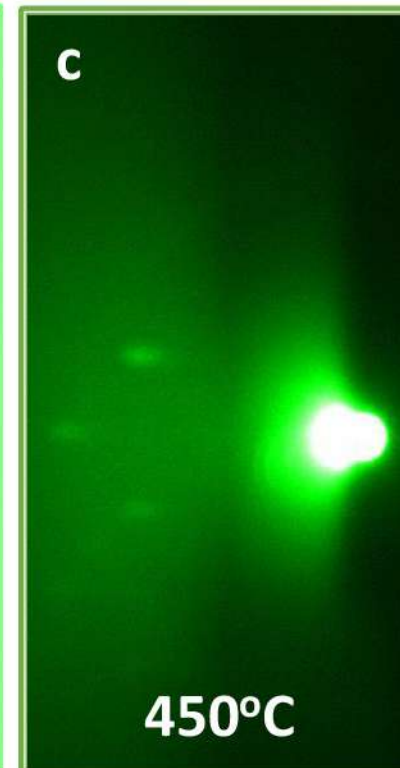
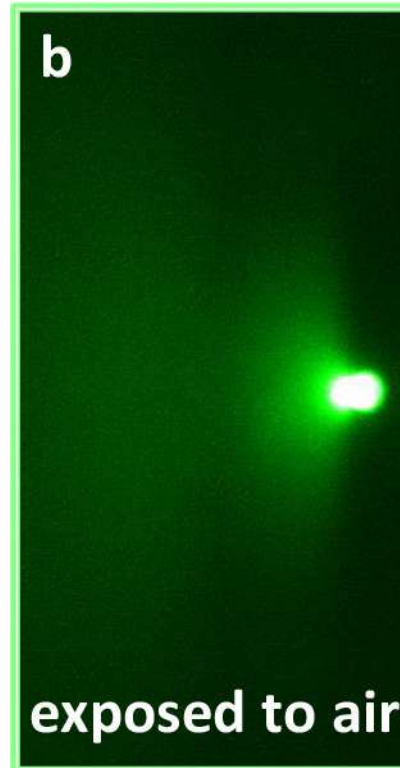
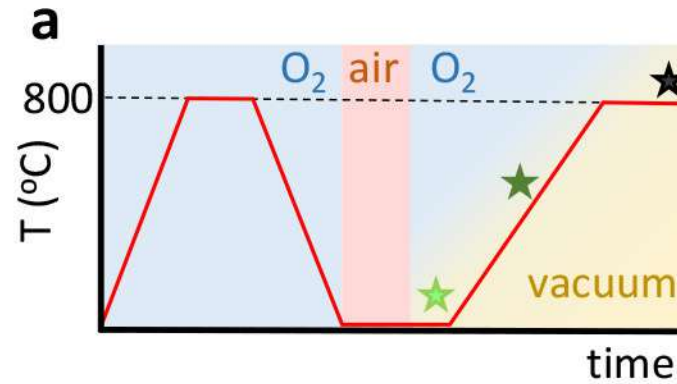


# Challenges : Oxide/SAO interface

Exposure to air/humidity

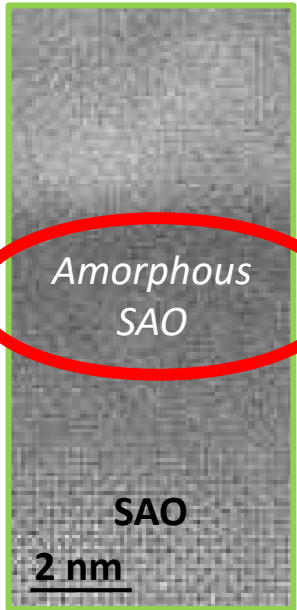


annealing in vacuum  
 →  
 Recover crystallinity

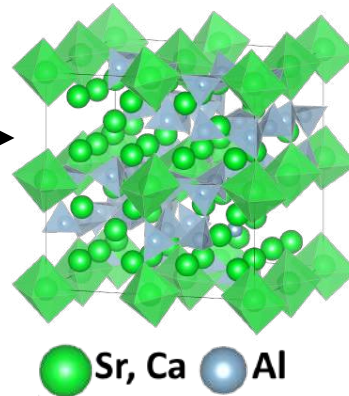
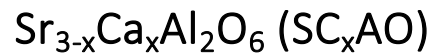


# Challenges : Oxide/SAO interface

## Cation engineering in SAO



Cation engineering the sacrificial layer:

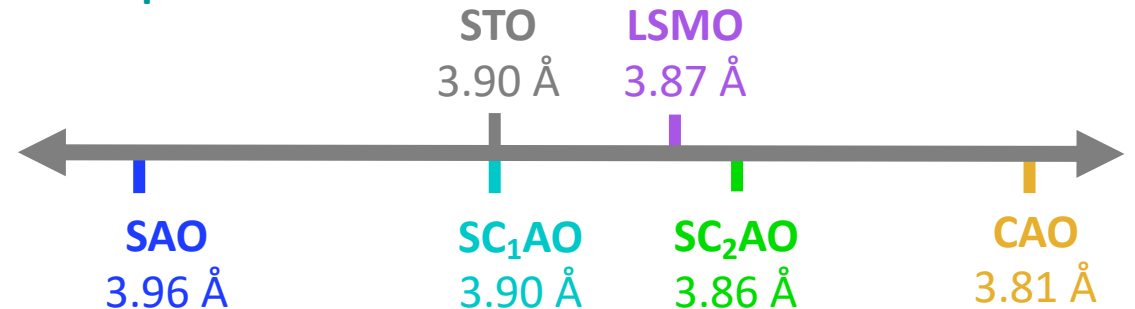


Cation	Electronegativity	Ionic radius (pm)	a (Å)	a/4 (Å)
Sr <sup>2+</sup>	0.95	132	15.84	3.96
Ca <sup>2+</sup>	1.00	114	15.26	3.81

### Reactivity with H<sub>2</sub>O

Al - O : covalent bond  
Sr, Ca - O : Ionic bond

### Lattice parameters





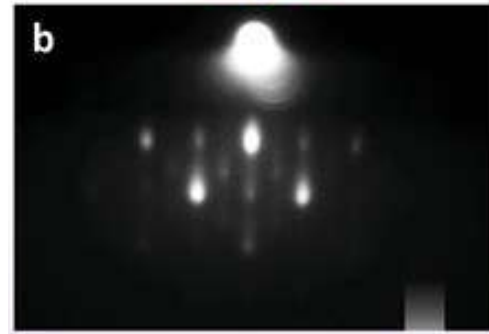
# Challenges : Stability of SAO

## Cation engineering

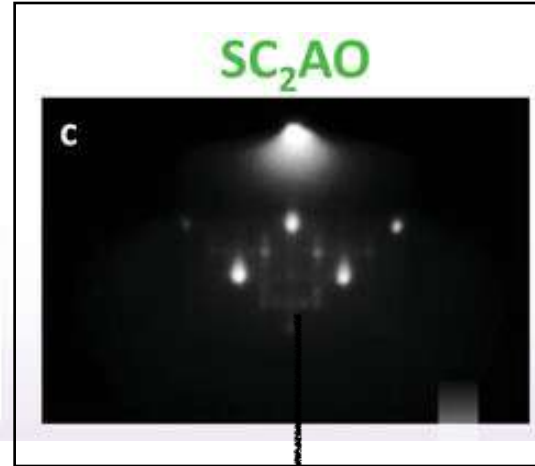
$SC_xAO$   
exposed to air



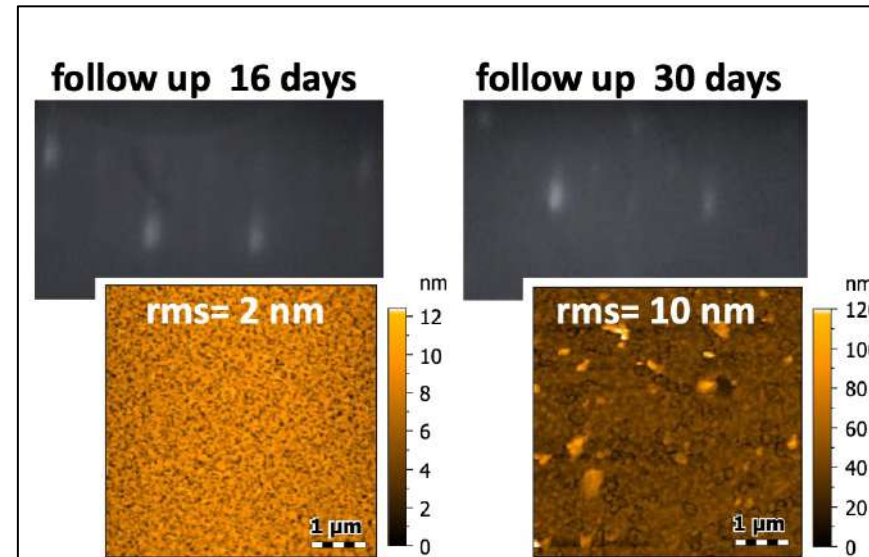
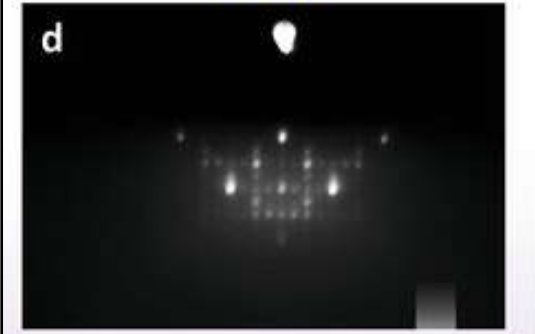
$SC_1AO$



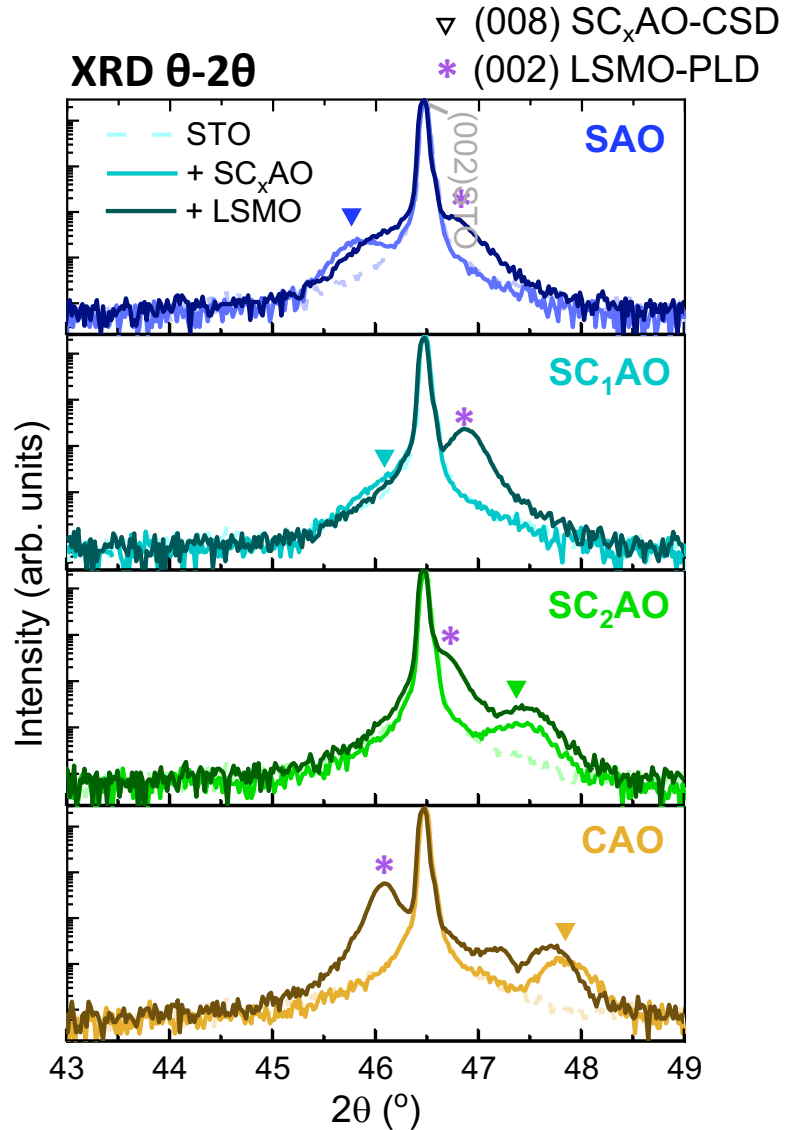
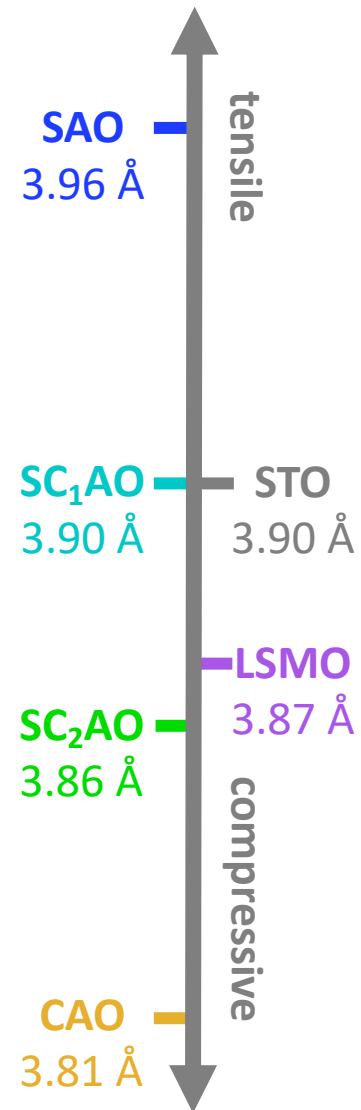
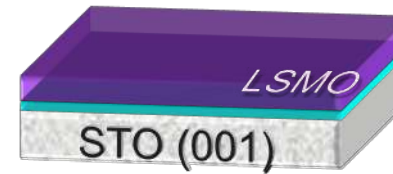
$SC_2AO$



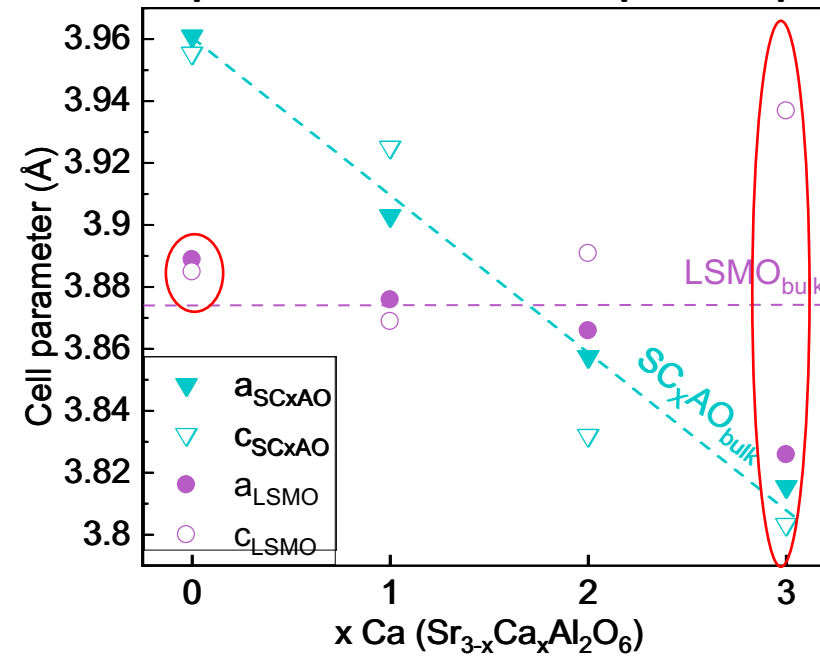
CAO



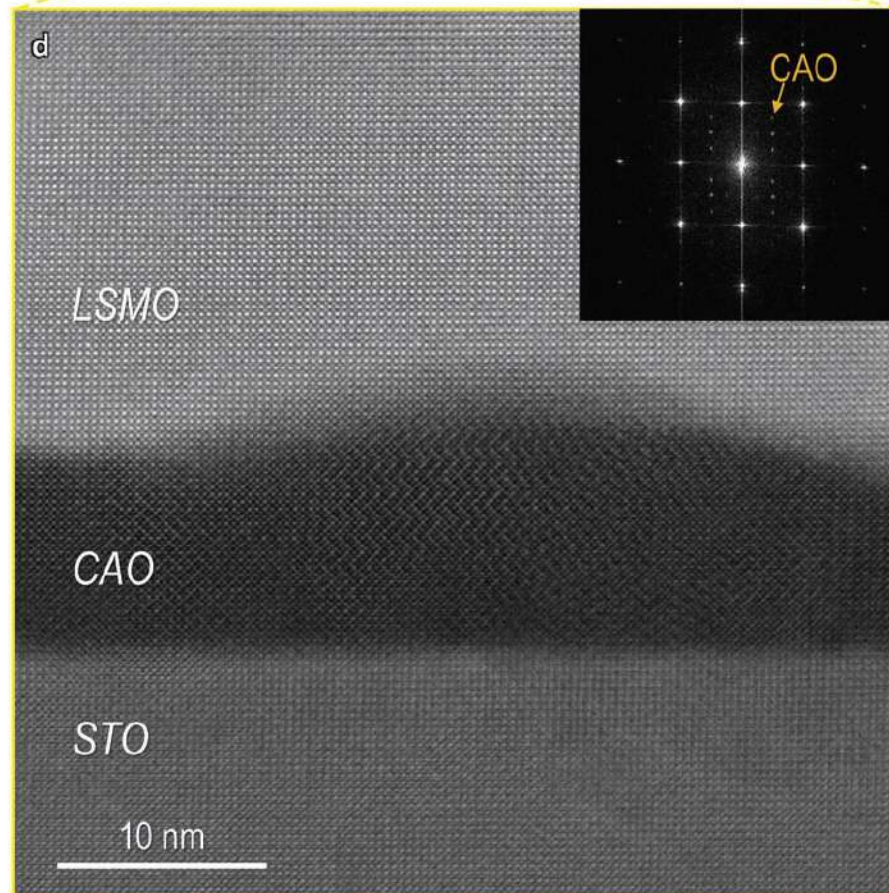
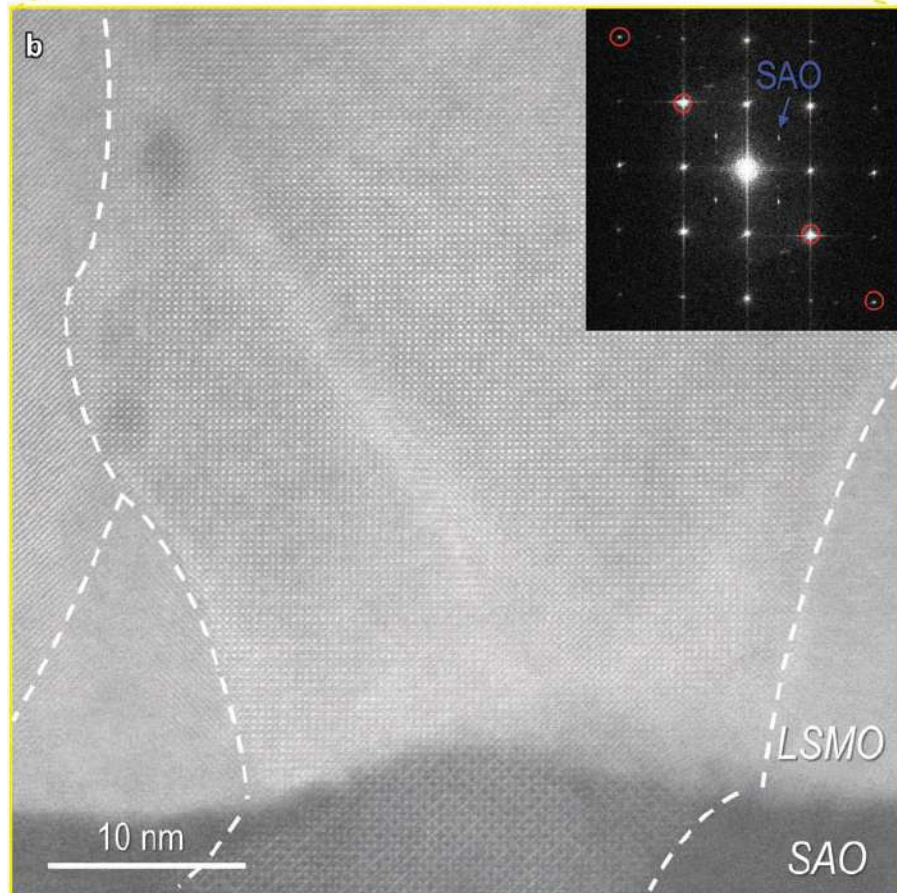
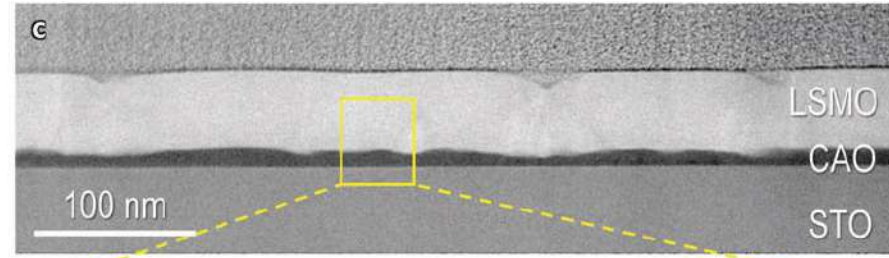
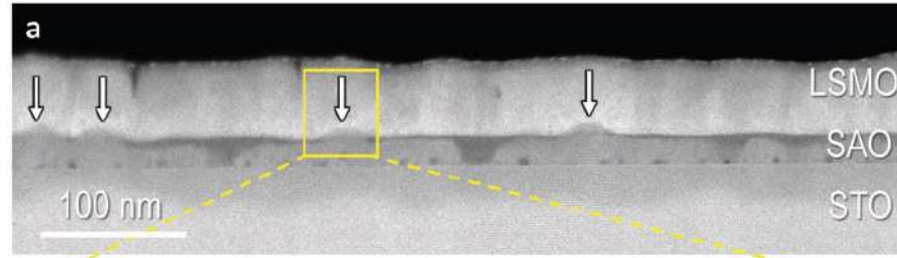
# LSMO/SC<sub>x</sub>AO//STO: XRD structure analysis



Lattice parameters from Reciprocal Space Map (RSM)

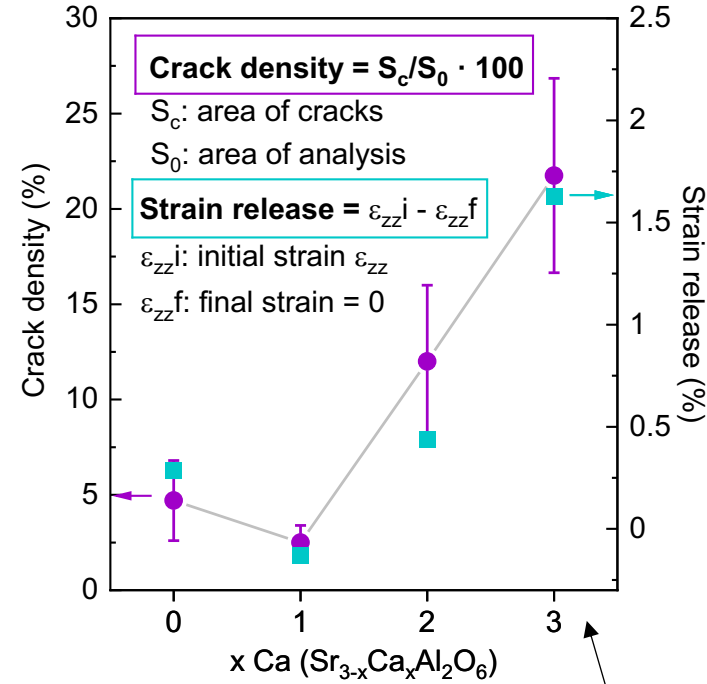
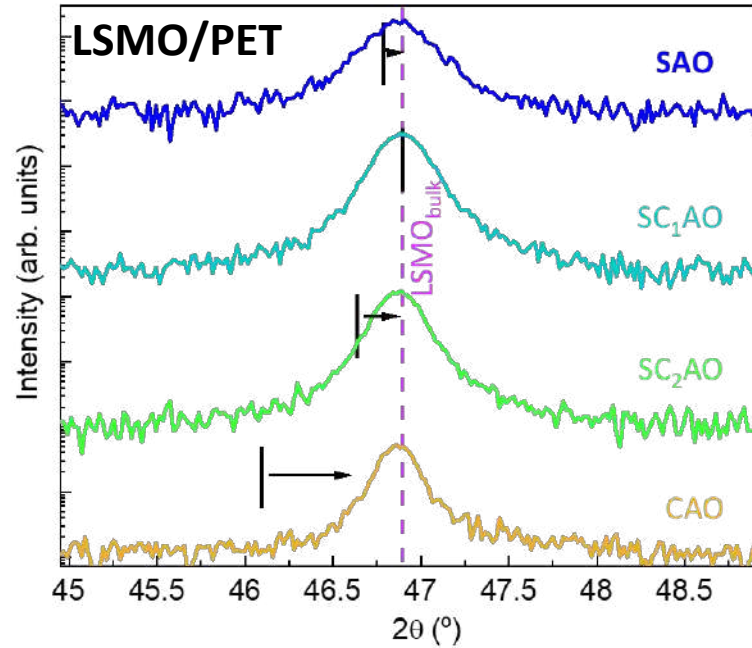
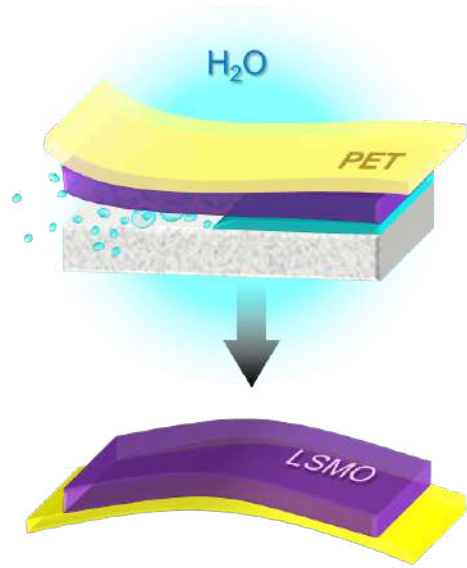


- Relaxed LSMO grown on SAO and SC<sub>1</sub>AO
- In-plane compressive strain for LSMO grown on SC<sub>2</sub>AO and CAO

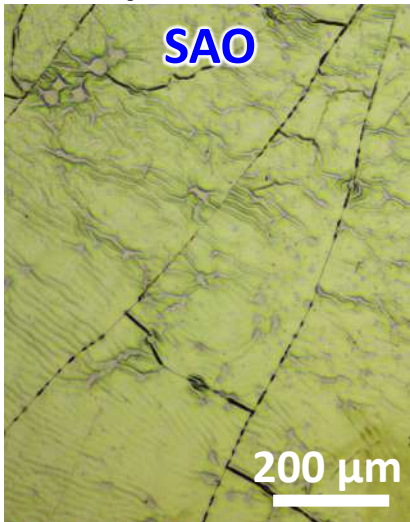


# LSMO membranes on PET: strain release

XRD  $\theta$ - $2\theta$

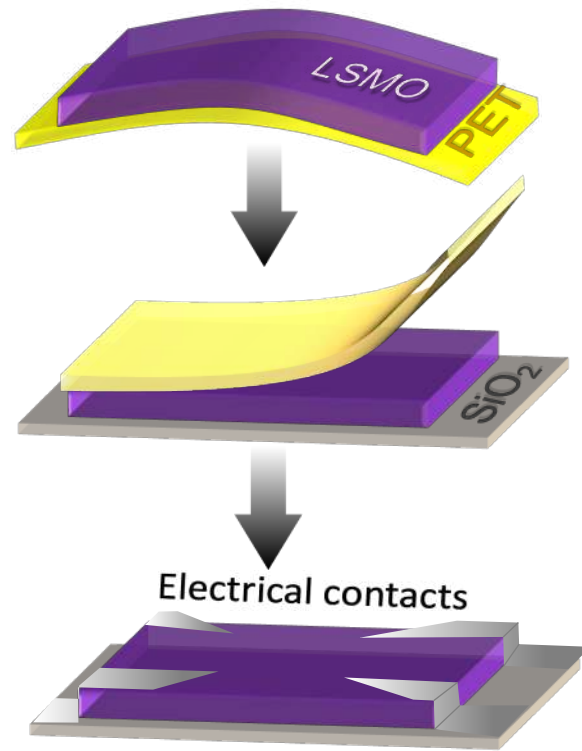


LSMO/PET

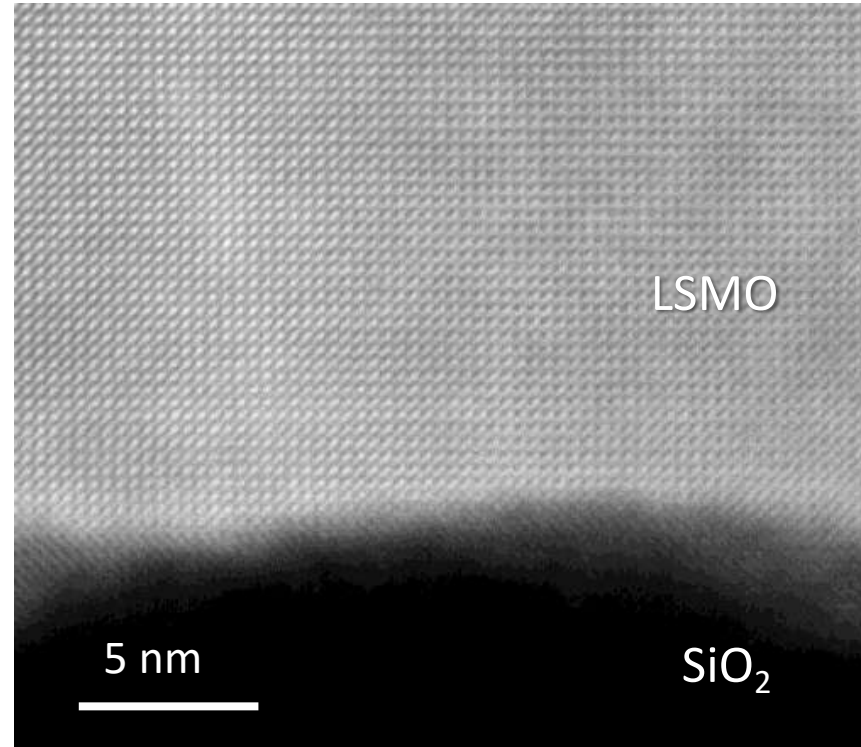


Crack density  $\propto$  strain released

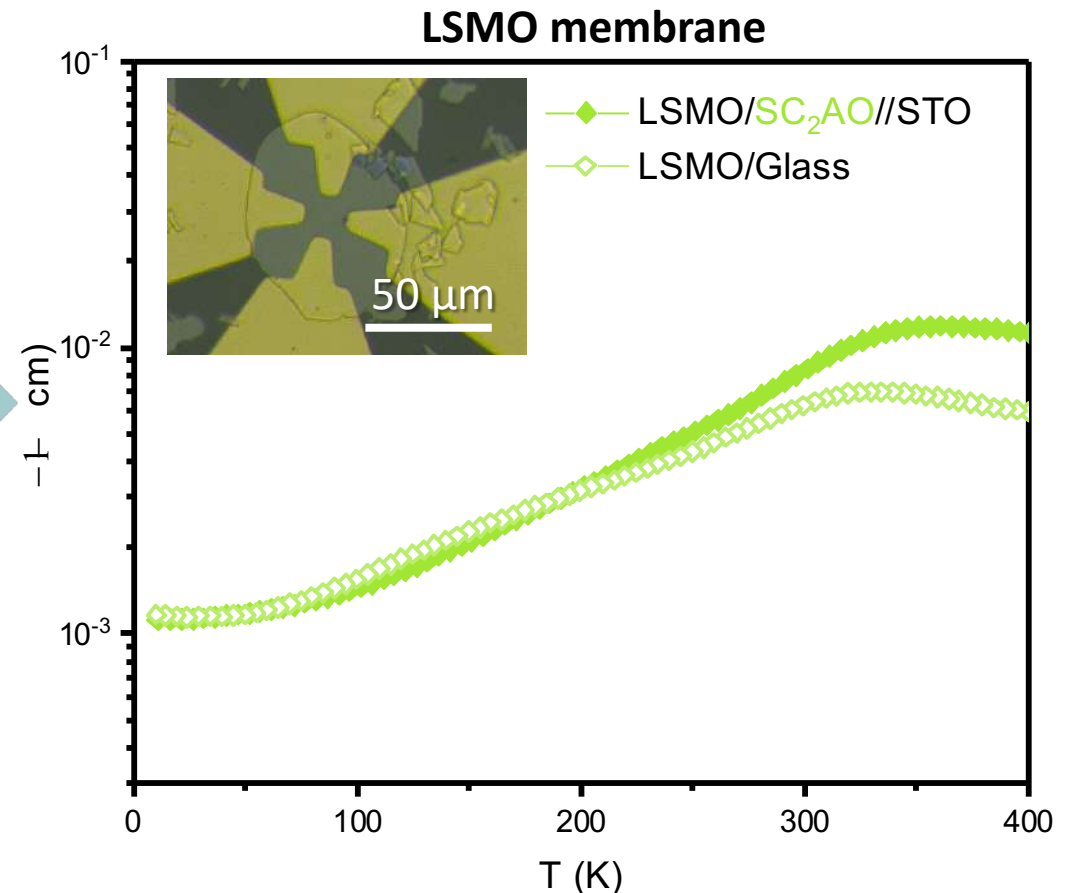
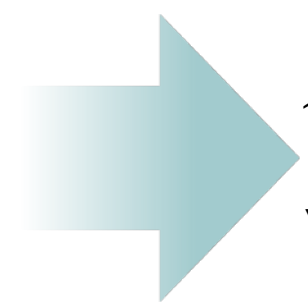
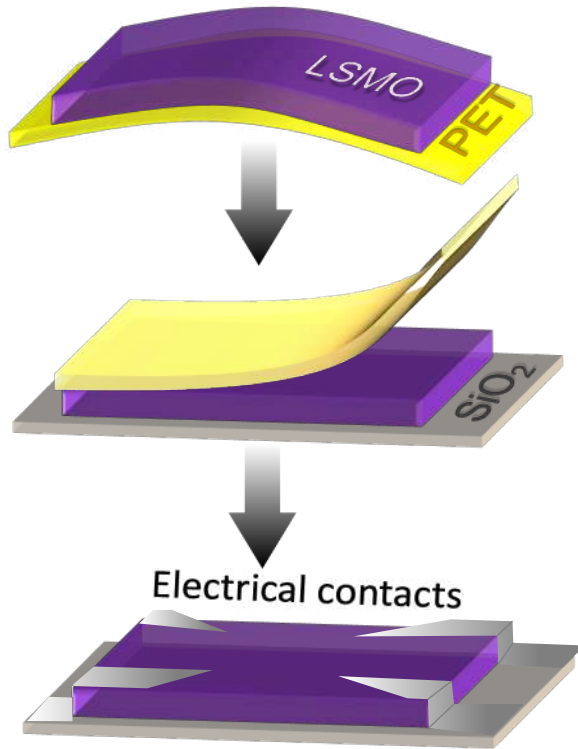
# LSMO electrical transport $\rho(T)$



STEM cross-section

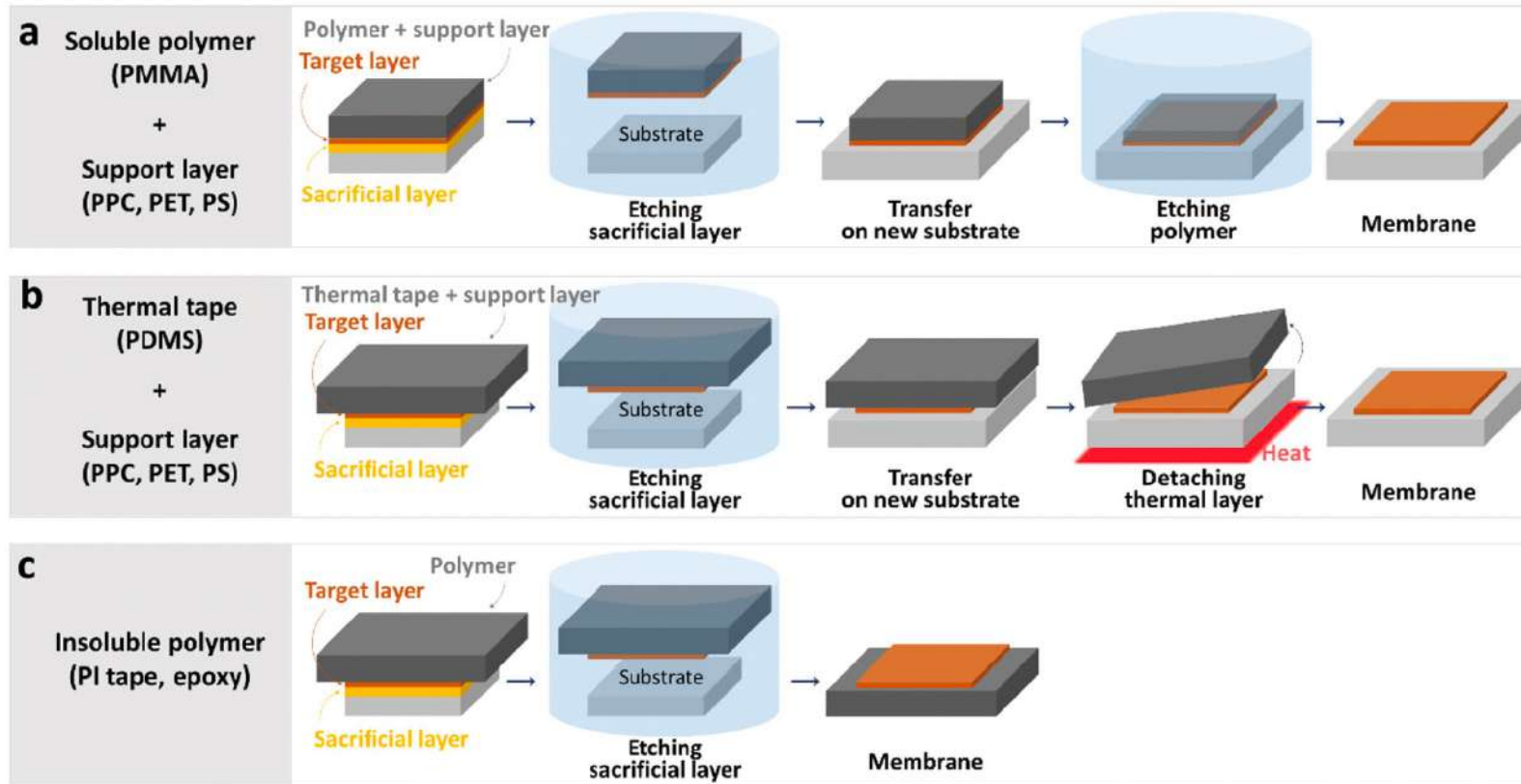


# LSMO electrical transport $\rho(T)$



- $\rho(T)$  of LSMO membrane presents small variations after release.

# Challenges : Exfoliation



*Solvents*

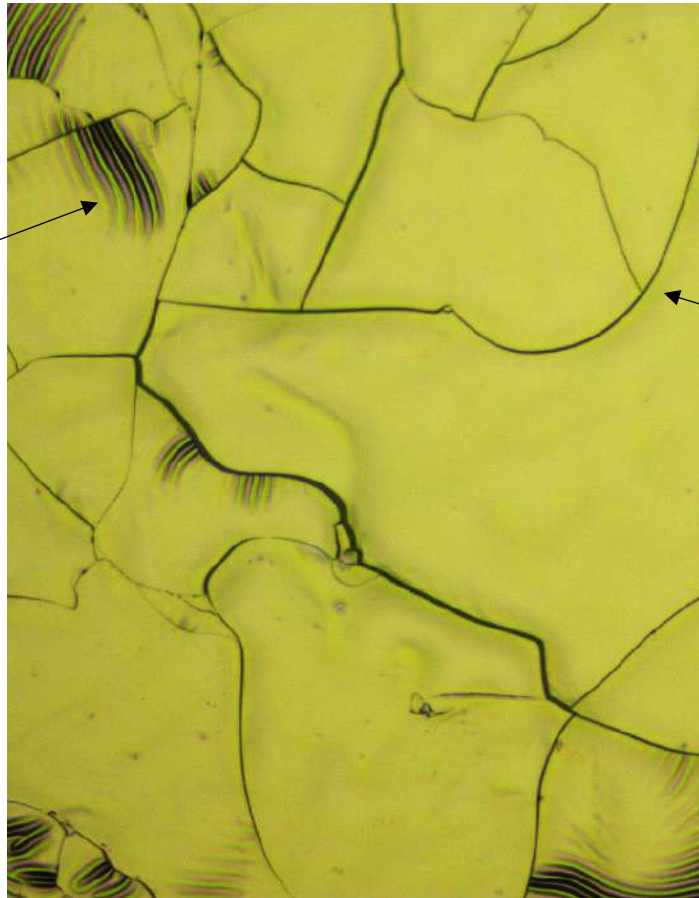
*Not easy to fully eliminate polymers  
Polymer cellulose (CAB) might be an alternative*

*Residues and cracks might be introduced*

*Cannot be transferred to another platform*

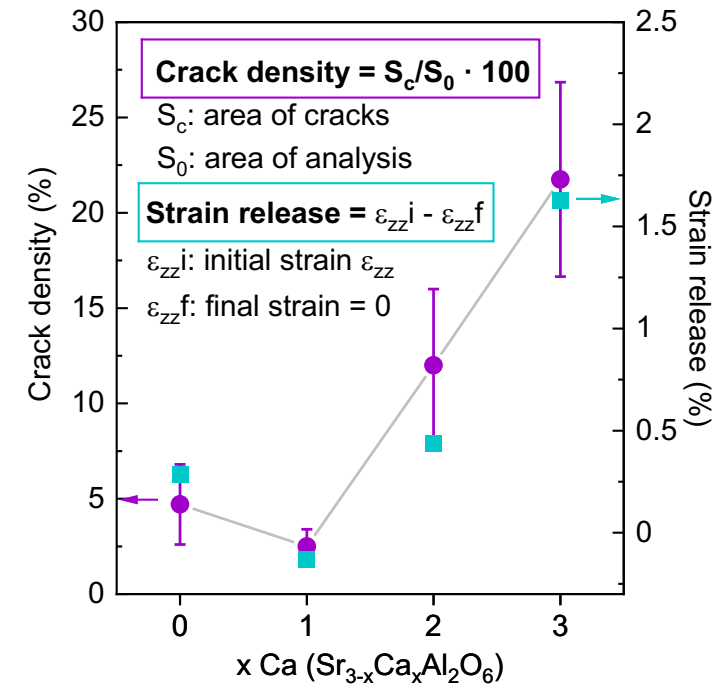
# Challenges : Exfoliation

Typical defects



wrinkles

cracks

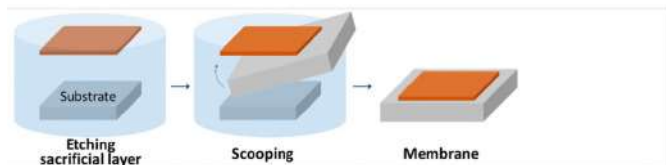


attributed to the stress during the releasing of the sup-port layer

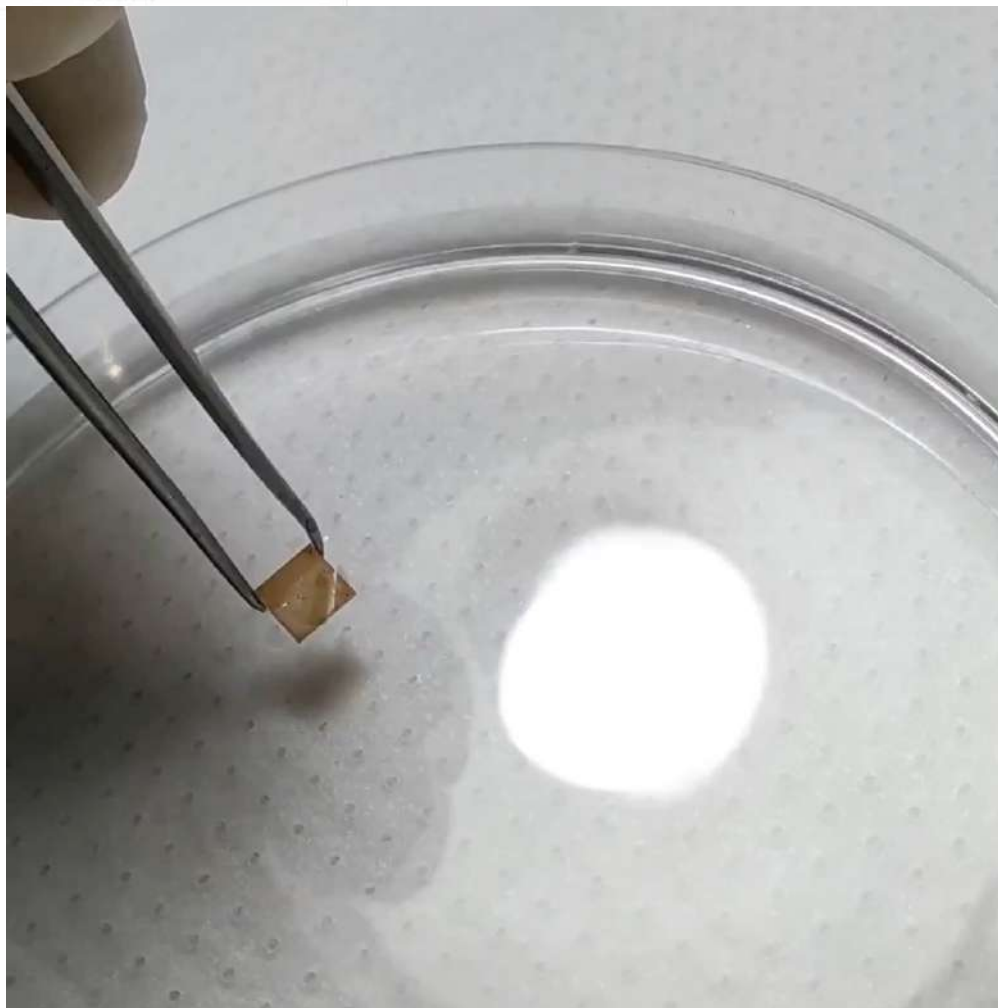


# Challenges : Exfoliation

Scooping



*Mechanical stability...*



- Why free-standing oxides
- Preparation of free-standing oxides
- Challenges in the preparation
- Outlook/ perspective

- Many new sacrificial layer composition and structures are yet to be developed (garnet and spinel not optimized)
- Understanding of how to transfer large defect-free membrane size is still needed in the short term
- Creation of novel electronic and photonic devices with superior performance: engineer properties and structure : strain-bending; artificial heterostructures, magic angle , 2D...
- Operando probes combined with theoretical studies to better elucidate the behavior of freestanding

# Acknowledgements

## CHEMOX TEAM-ICMAB



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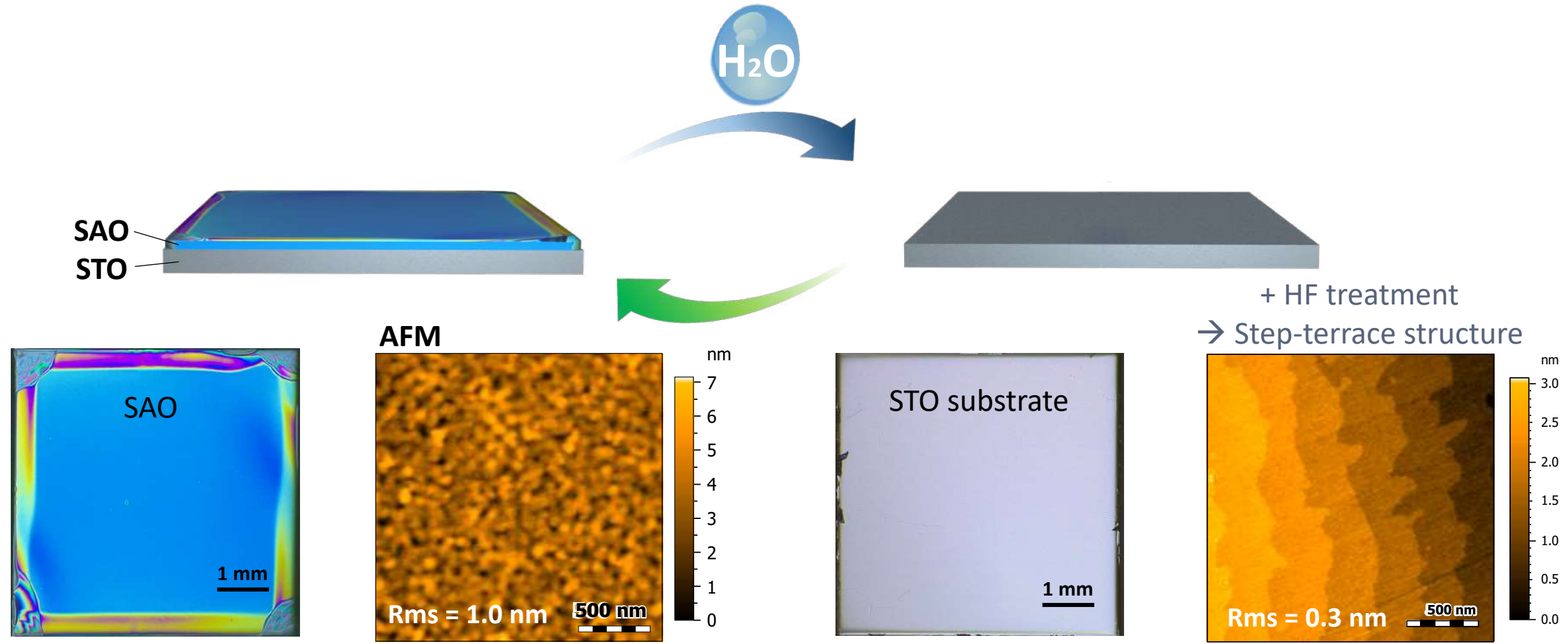
D. Zanders,  
Prof. Dr. A. Devi  
for precursor synthesis  
@ Bochum University,  
Germany

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# Characteristics of optimized SAO//STO epitaxial film



✓ Epitaxy and smooth surface

- ✓ Dissolved in H<sub>2</sub>O
- ✓ Substrate recycling